

A thick blue curved line that starts on the left side of the slide and arches towards the right side.

Using Computational Thermochemistry in Metallurgical Plants

Contents



- General issues concerning the use of FACTSAGE
- Several practical examples

- General FACTSAGE approach:
 - Fit full (multi-component) system
 - Present results in phase diagrams etc.

- Industry wants:
 - Focus on limited compositional ranges
 - Results in terms of grade/degree of conversion

More process minded vs. reaction minded

 - Mass and heat balance

General



- Metallurgical systems
 - Multi-component systems
 - Multi-phase “but not that many”

- Possible and risky use of FACTSAGE:
- Enter your chemical elements:
- Fe, Si, Al, Mg, Ca, Ti, P, Mn, C, O, H (multi-component)
- Only the default databases etc.
- Pure condensed phases ('many' multi-phase)

General



- At least investigate a sample of the material:
 - 1) XRD phase identification
 - 2) Optical microscopy
 - 3) Microprobe analysis / EDS

Examples

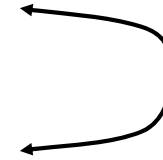


- Blast furnace slag (liquidus temperature)
- Converter slag (3 times): Lining attack (chemical)
Liquidus temperature
Free CaO
- Slag washing (lining protection; physical)
- Sinter process (prediction of phase composition)

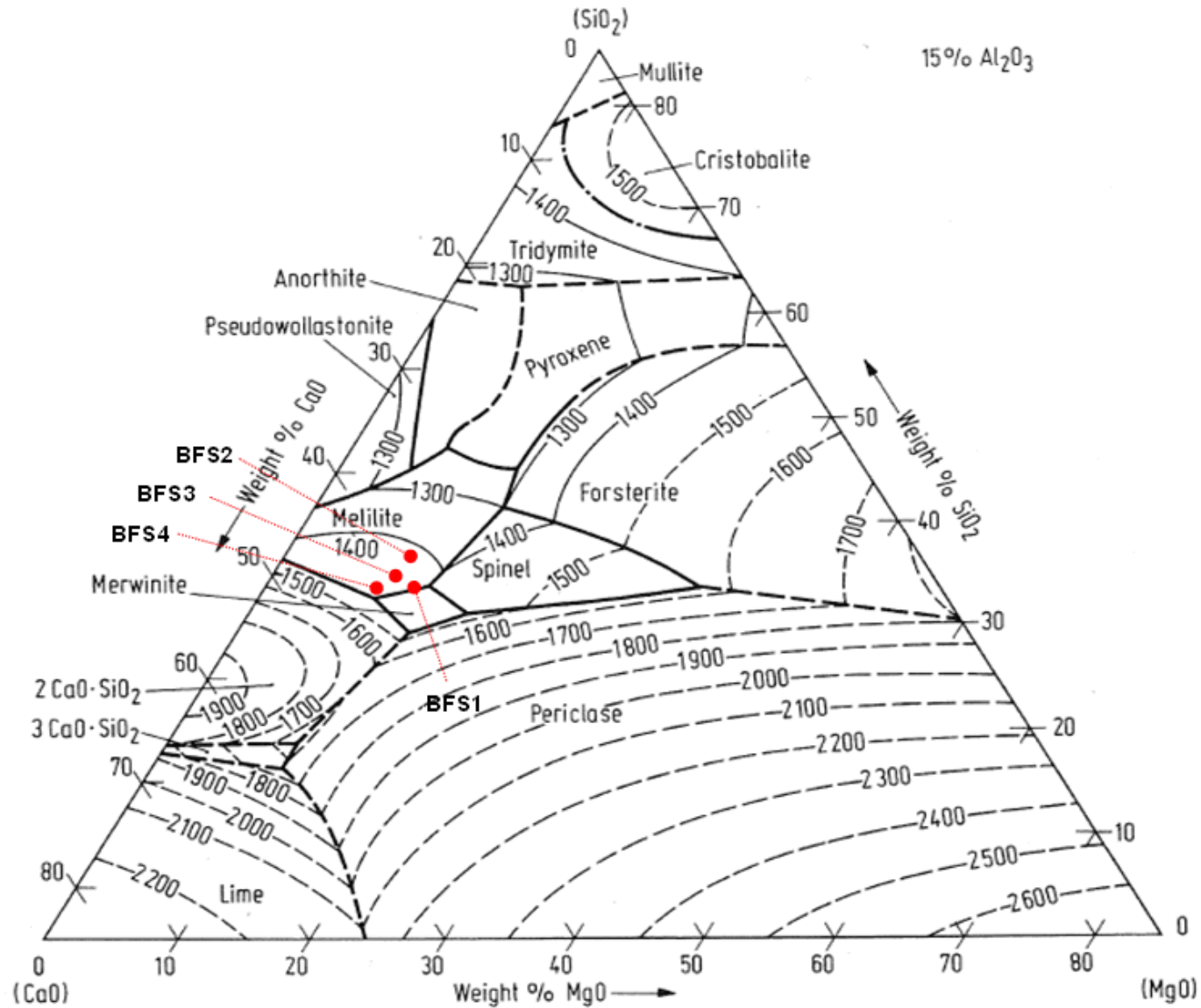
Blast furnace slag



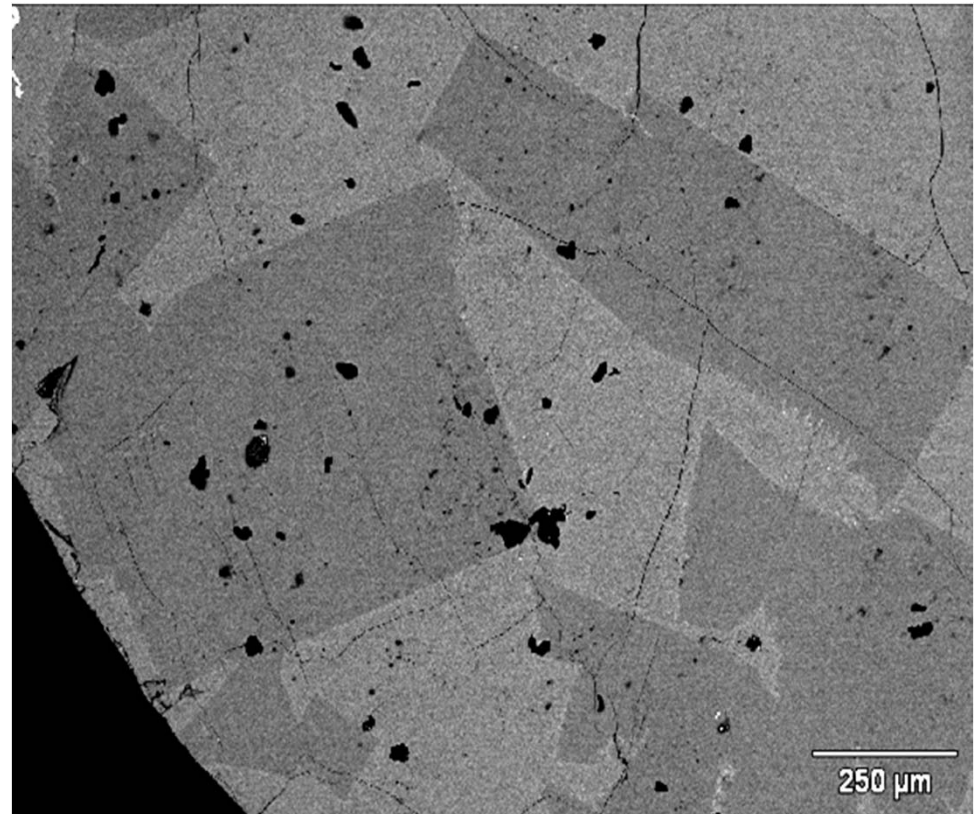
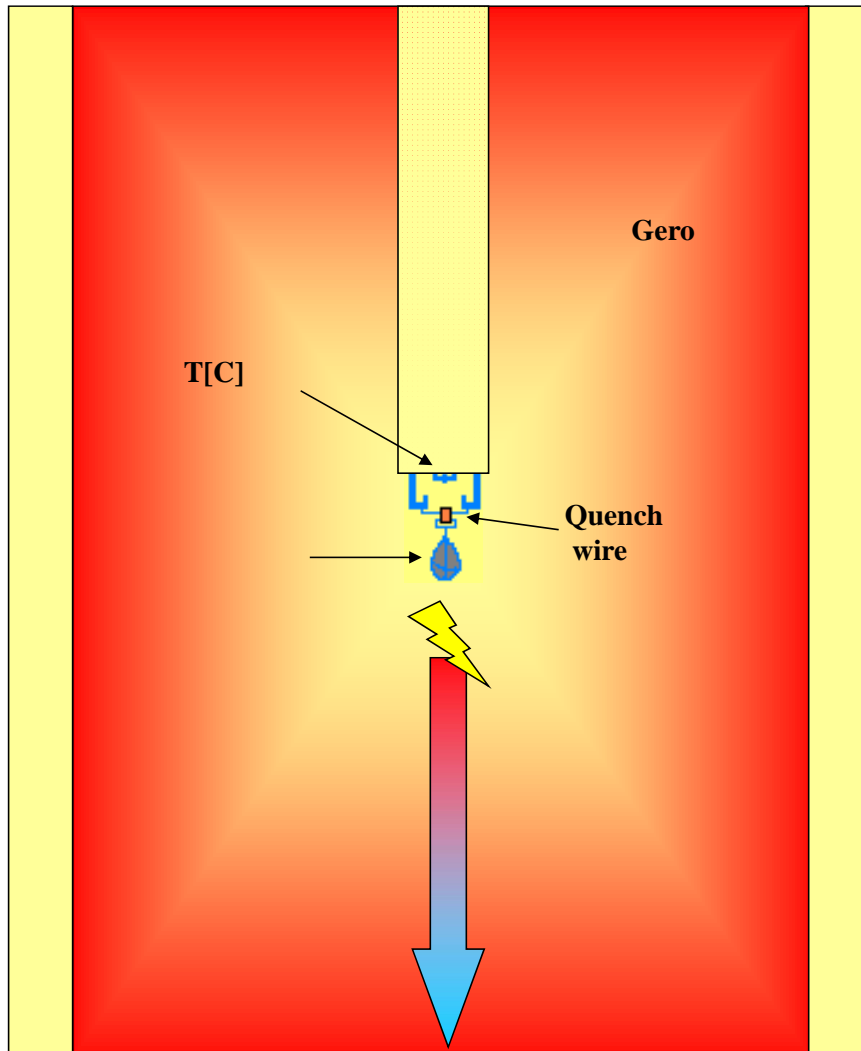
- Ex iron ore
- FeO nothing left
- SiO₂
- CaO added to neutralise the silica
- Al₂O₃
- MgO added for refractory protection



Blast furnace slag



Blast furnace slag

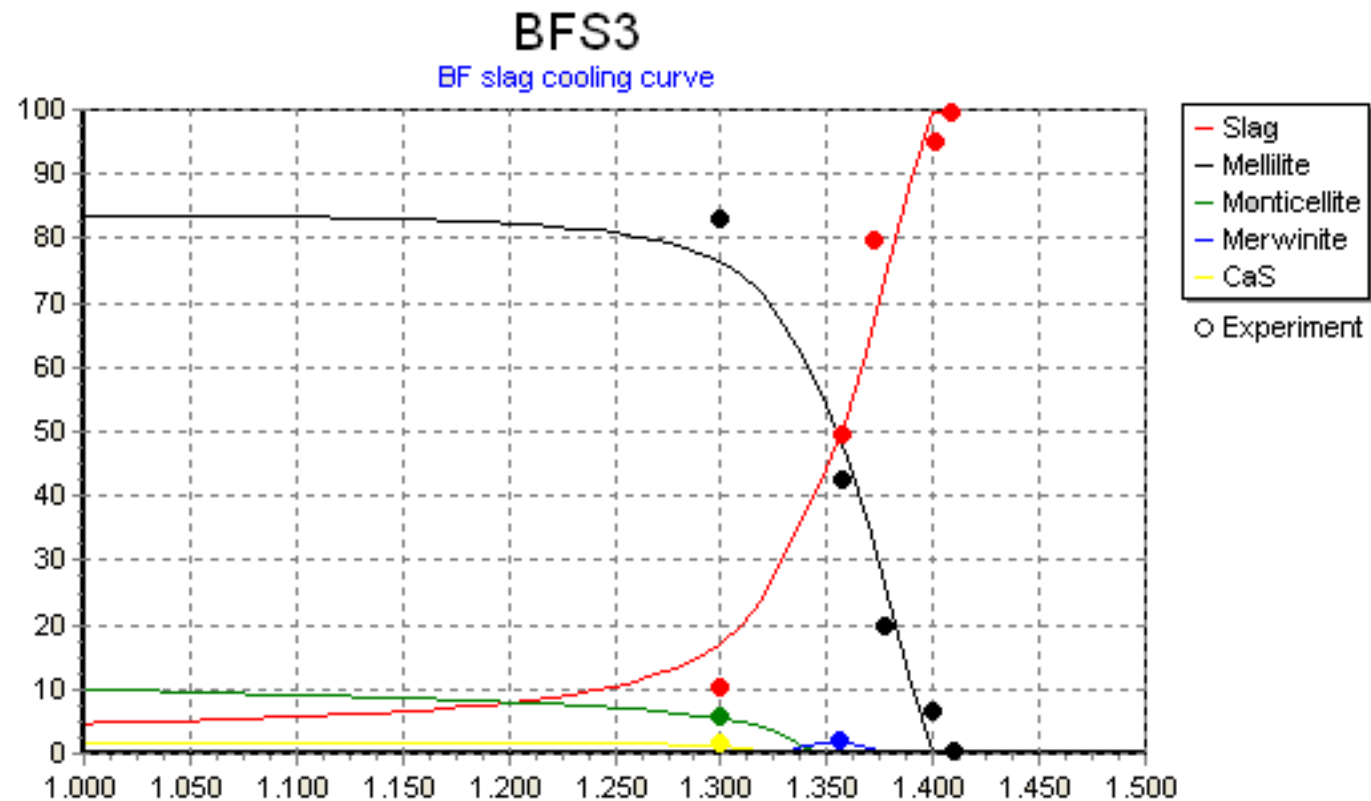


Blast furnace slag



Mass fraction of crystals observed?

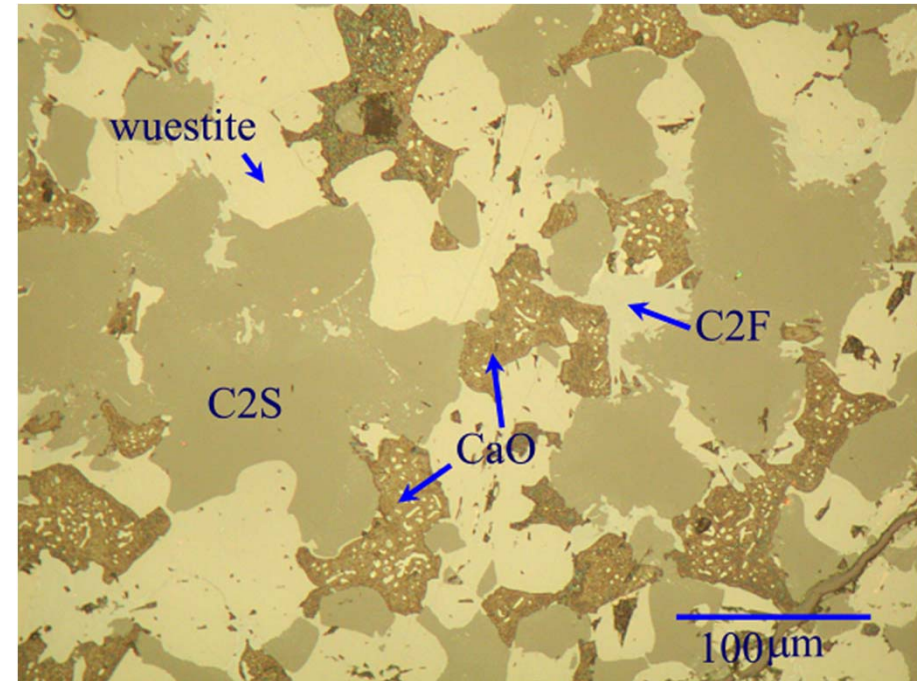
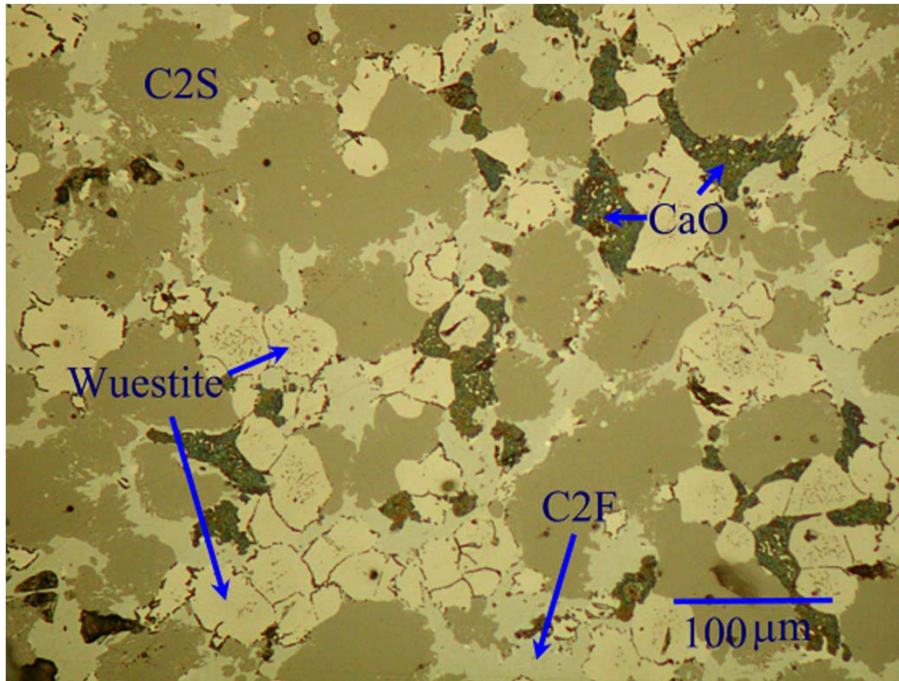
$100\% \cdot \text{Al}_2\text{O}_3\%$ in slag = $x \cdot \text{Al}_2\text{O}_3$ in crystal + $(100-x) \cdot \text{Al}_2\text{O}_3$ in liquid
For all components optimise



Blast furnace slag

- Need some minor adaptations of G(phase) or allow some systematic error which is NOT accepted
- Use full chemical analysis
- Prepare cst file within FACTSAGE
- Write dedicated software (e.g. in Delphi)
- Use ChemApp.dll (with saved cst file)

Converter slag



Components (9): FeO, MnO, MgO, CaO, SiO₂, P₂O₅, Al₂O₃, Fe₂O₃, TiO₂

Converter slag

• Components:

- FeO
- MnO
- MgO
- CaO
- SiO₂
- P₂O₅
- Al₂O₃
- Fe₂O₃
- TiO₂

Phases:

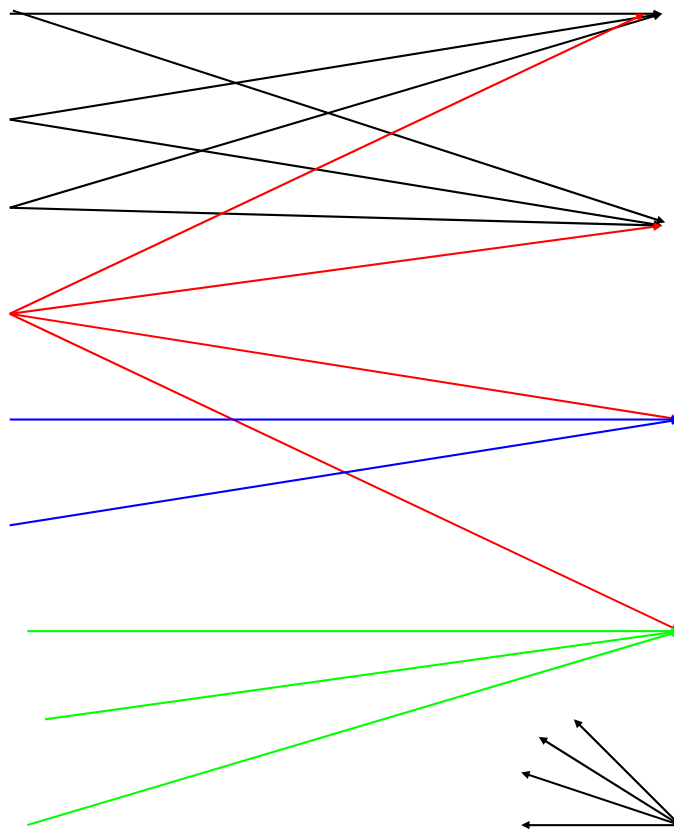
MgO wustite (MgO,FeO,MnO,CaO)

free CaO (CaO,MnO,FeO,MgO)

silicate (Ca₂(Si,P)O₄)

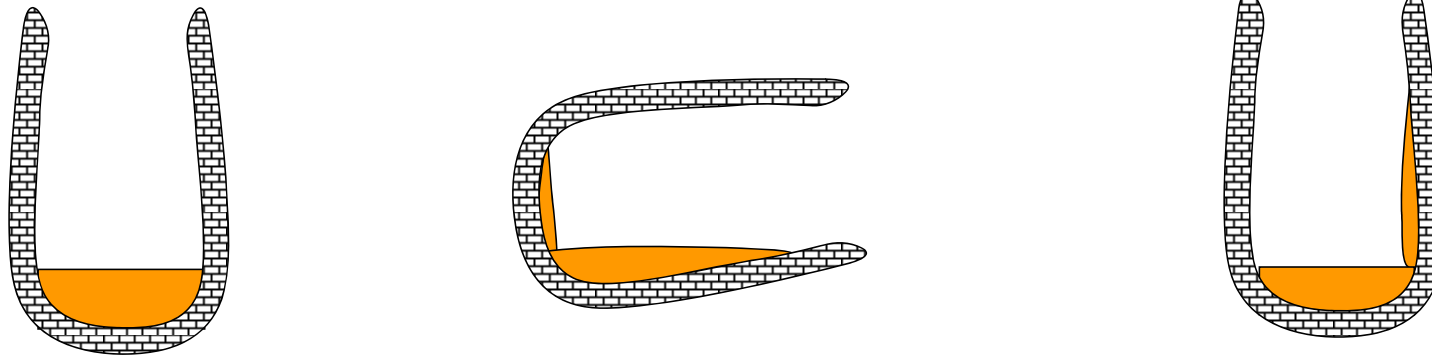
ferrite (Ca₂(Fe,Al,Ti)₂O₅)

slag (liquid)



Slag washing

- Purpose: protective slag layer on top of refractory



- Enhance life-time converter

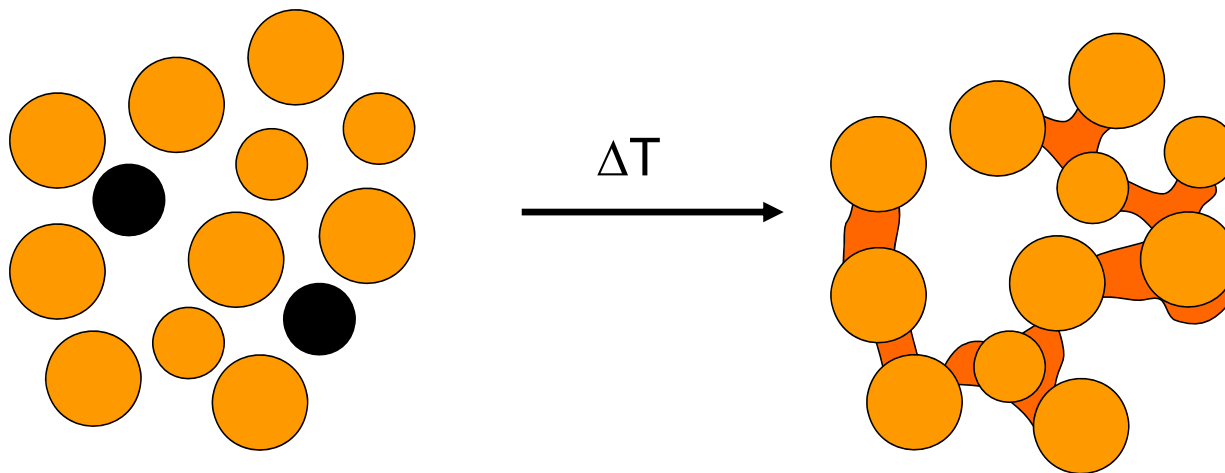
Key issue: viscosity S/L ratio

Slag washing

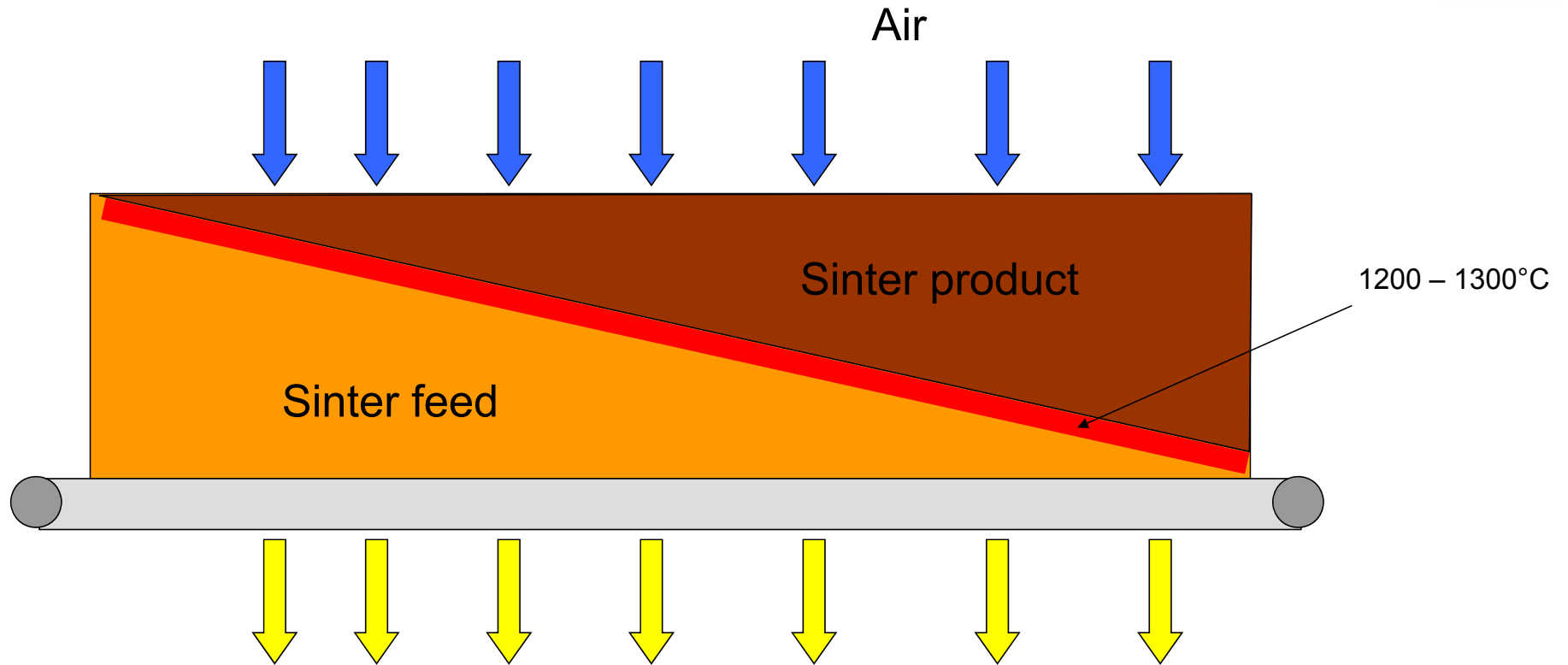
- Simple control: Temperature desired S/L ratio
- Add cool slag to decrease T to the right level
- Heat balance calculation (data ex FACTSAGE)

Sinter process

- Input preparation for blast furnace
- Dust/ fine ore Lumps



Sinter process



Iron ore: Fe_2O_3
 SiO_2
 Al_2O_3



CaO
C additives

Sinter process

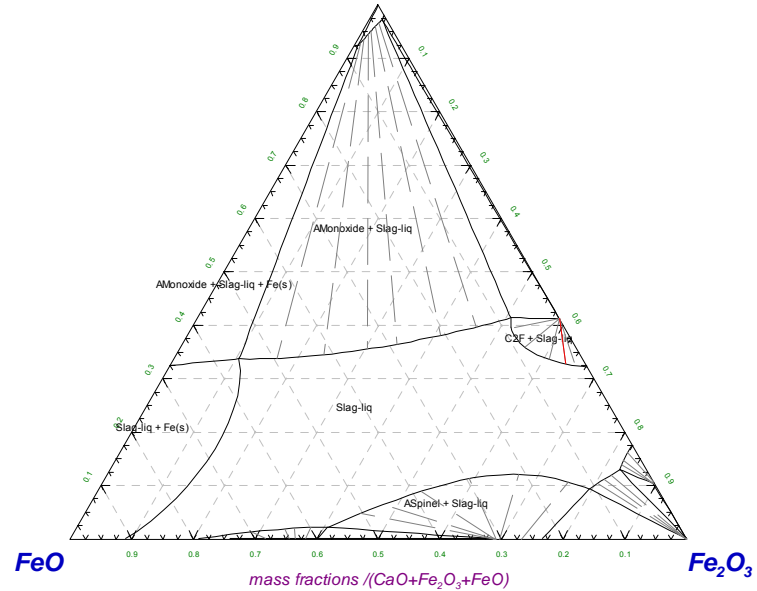
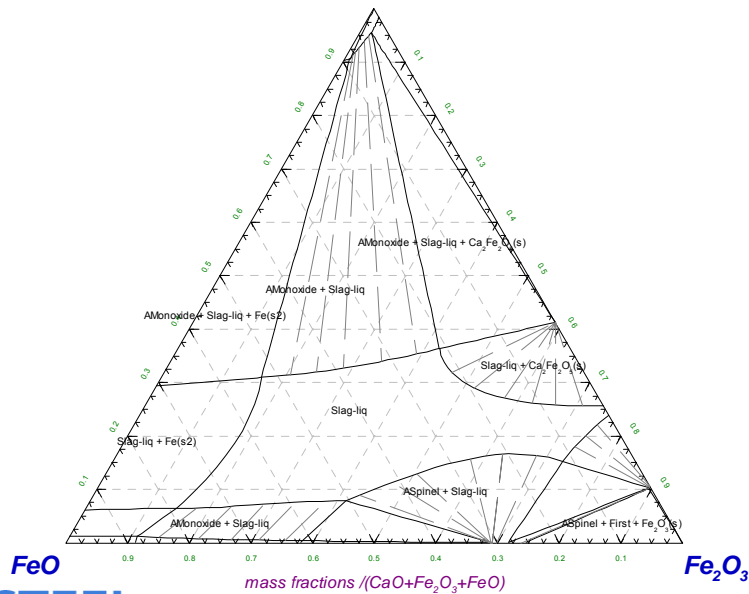
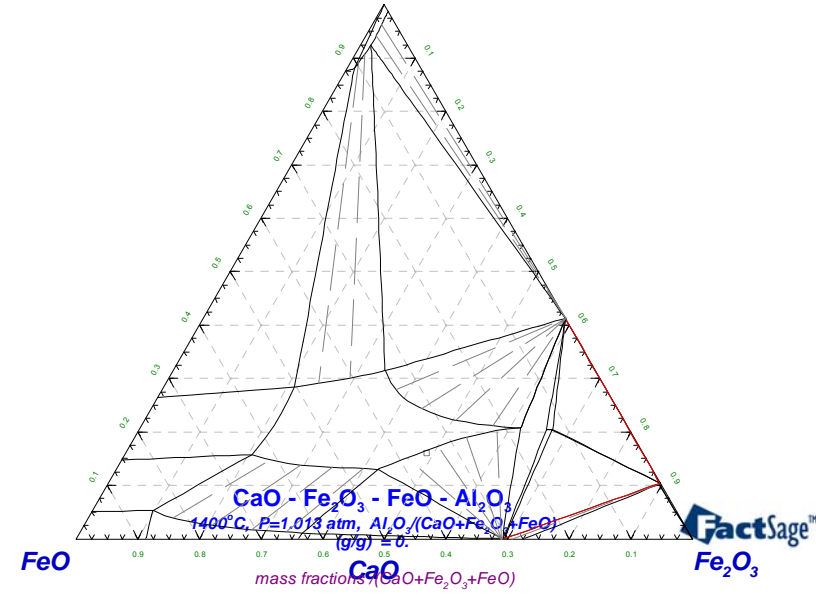
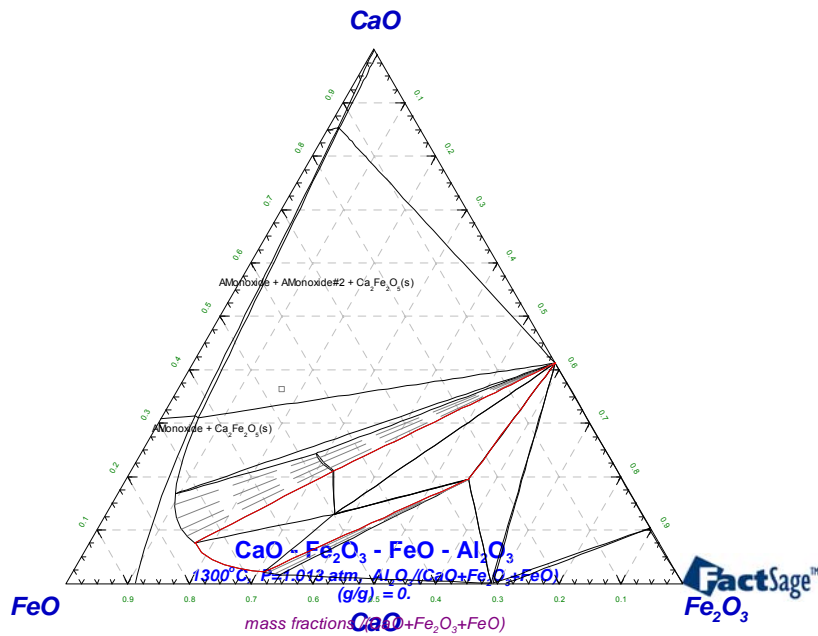


CaO - Fe₂O₃ - FeO - Al₂O₃
 1100°C, P=1.013 atm, Al₂O₃/(CaO+Fe₂O₃+FeO)
 (g/g) = 0.



CaO - Fe₂O₃ - FeO - Al₂O₃
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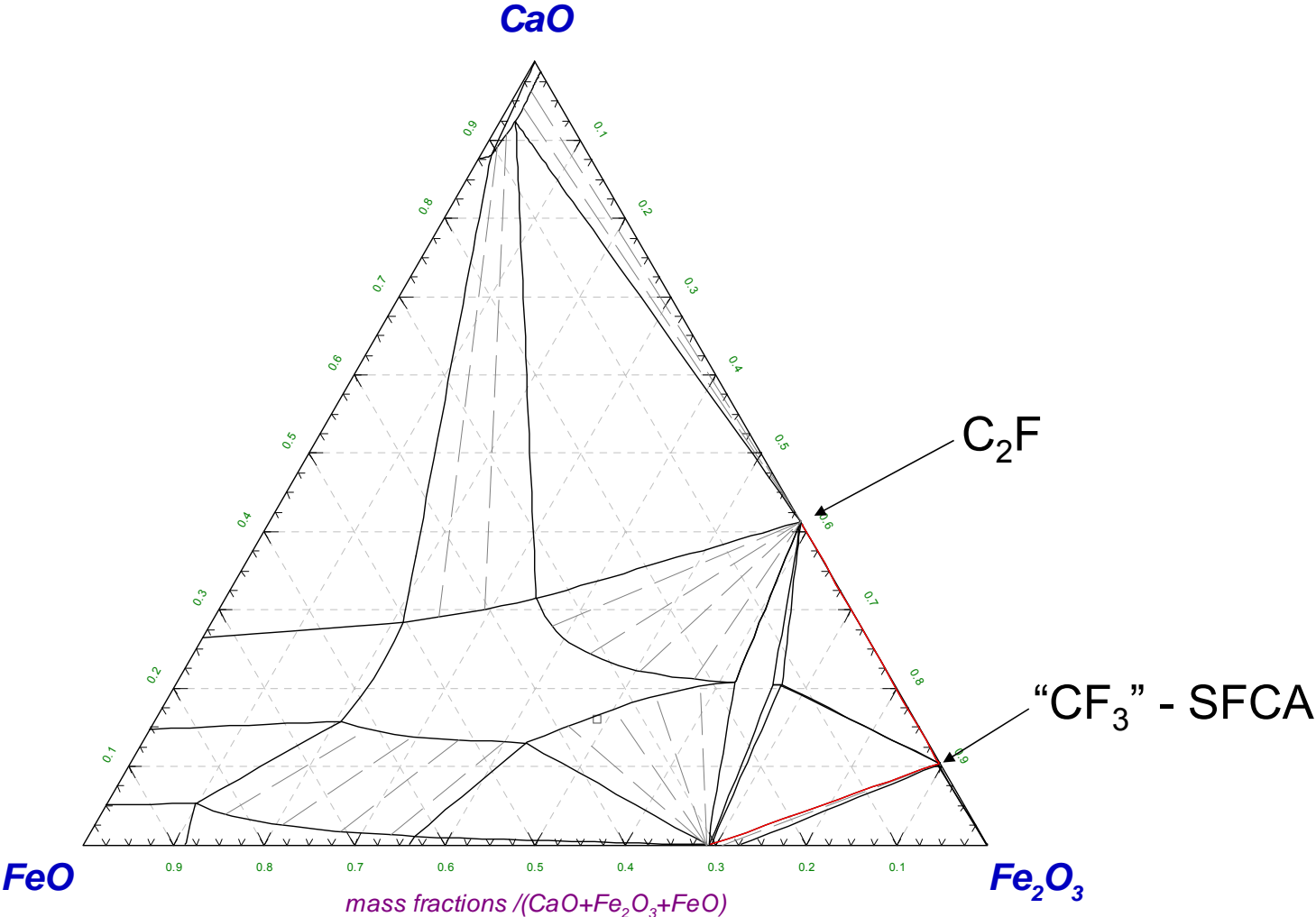
CaO



Sinter process



CaO - Fe₂O₃ - FeO - Al₂O₃
 1200°C, P=1.013 atm, Al₂O₃/(CaO+Fe₂O₃+FeO)
 (g/g) = 0.



How to model



Thermodynamics == FACT

NOT by using the interactive FactSage interface

Step 1. Organise data needed in FactSage !

Step 2. Make sub-database in FactSage !

This case: slag, metal, gas phase

Step 3. Use ChemApp.dll instead