

CALCULATION AND INTERPRETATION OF PROJECTIONS OF THE TEMPERATURE OF FIRST MELTING

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Outline

- Solidus projection (6.3)
- First-melting-temperature projection (6.4)

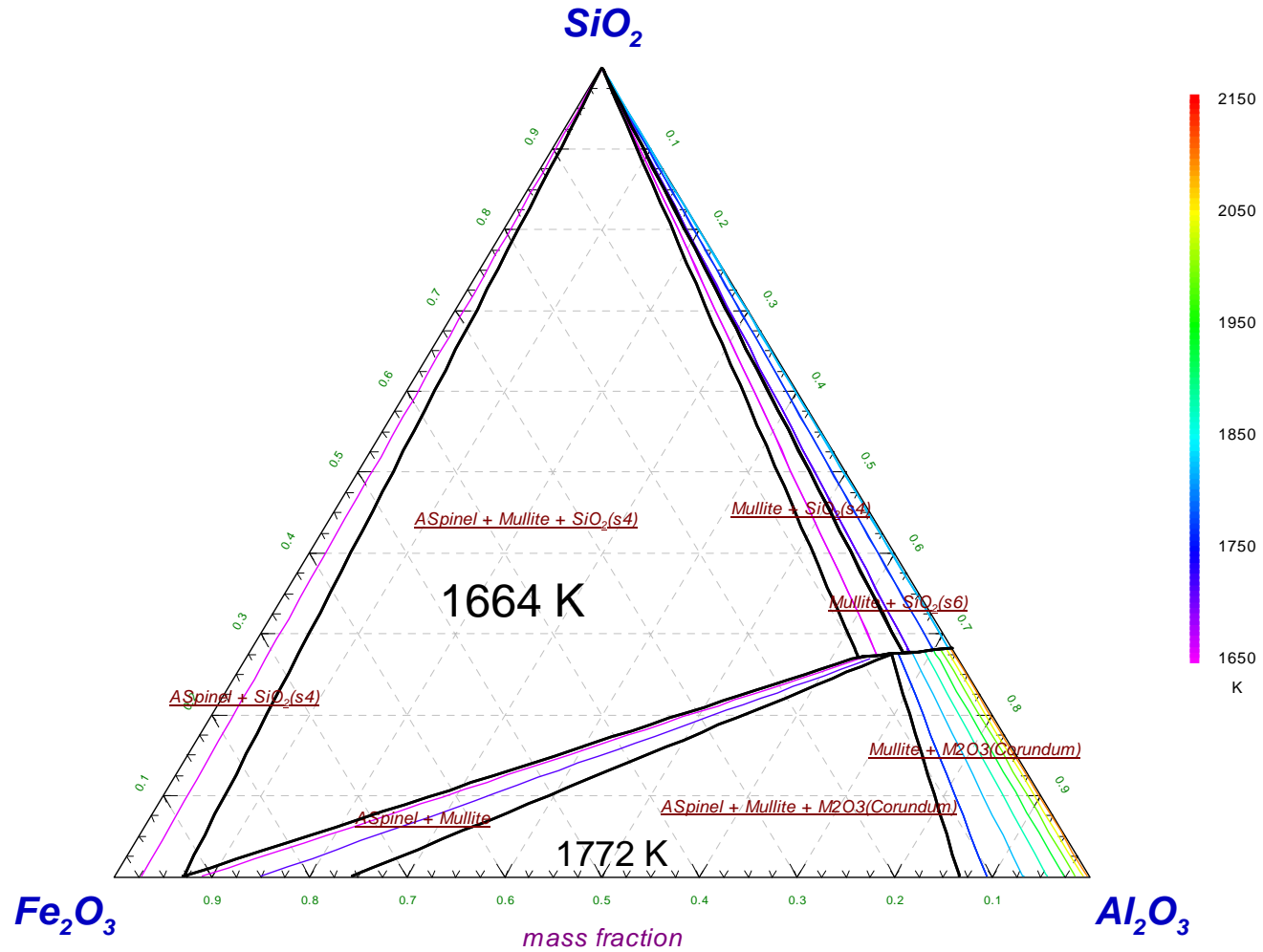


Outline

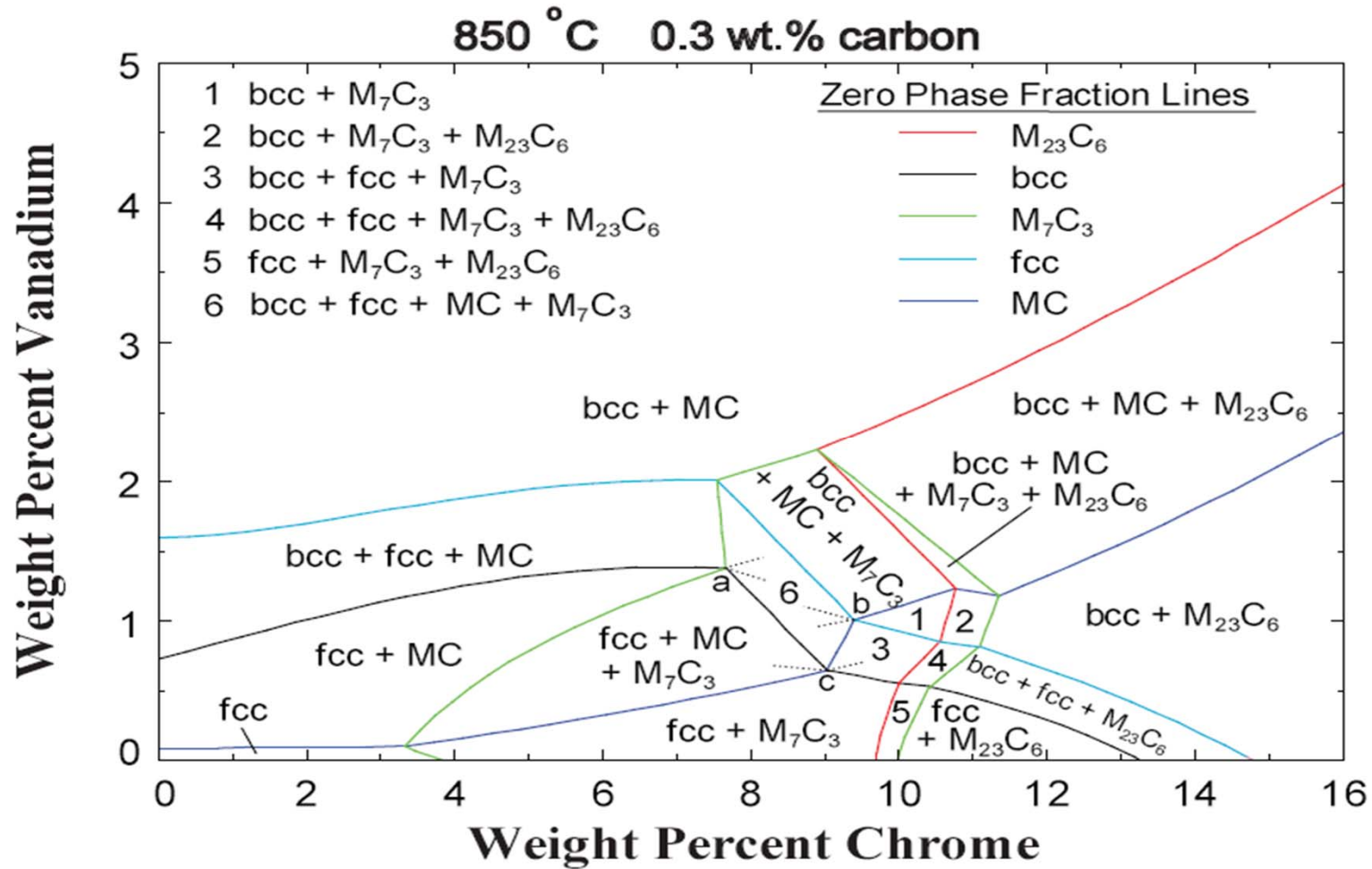
- Solidus projection (6.3)
- First-melting-temperature projection (6.4)



$\text{Fe}_2\text{O}_3 - \text{Al}_2\text{O}_3 - \text{SiO}_2 - \text{O}_2$
Projection - formation (ASlag-liq), $p(\text{O}_2) = 0.21 \text{ bar}, 1 \text{ bar}$



Fe-Cr-V-C isothermal section illustrating Zero-Phase-Fraction (ZPF) lines

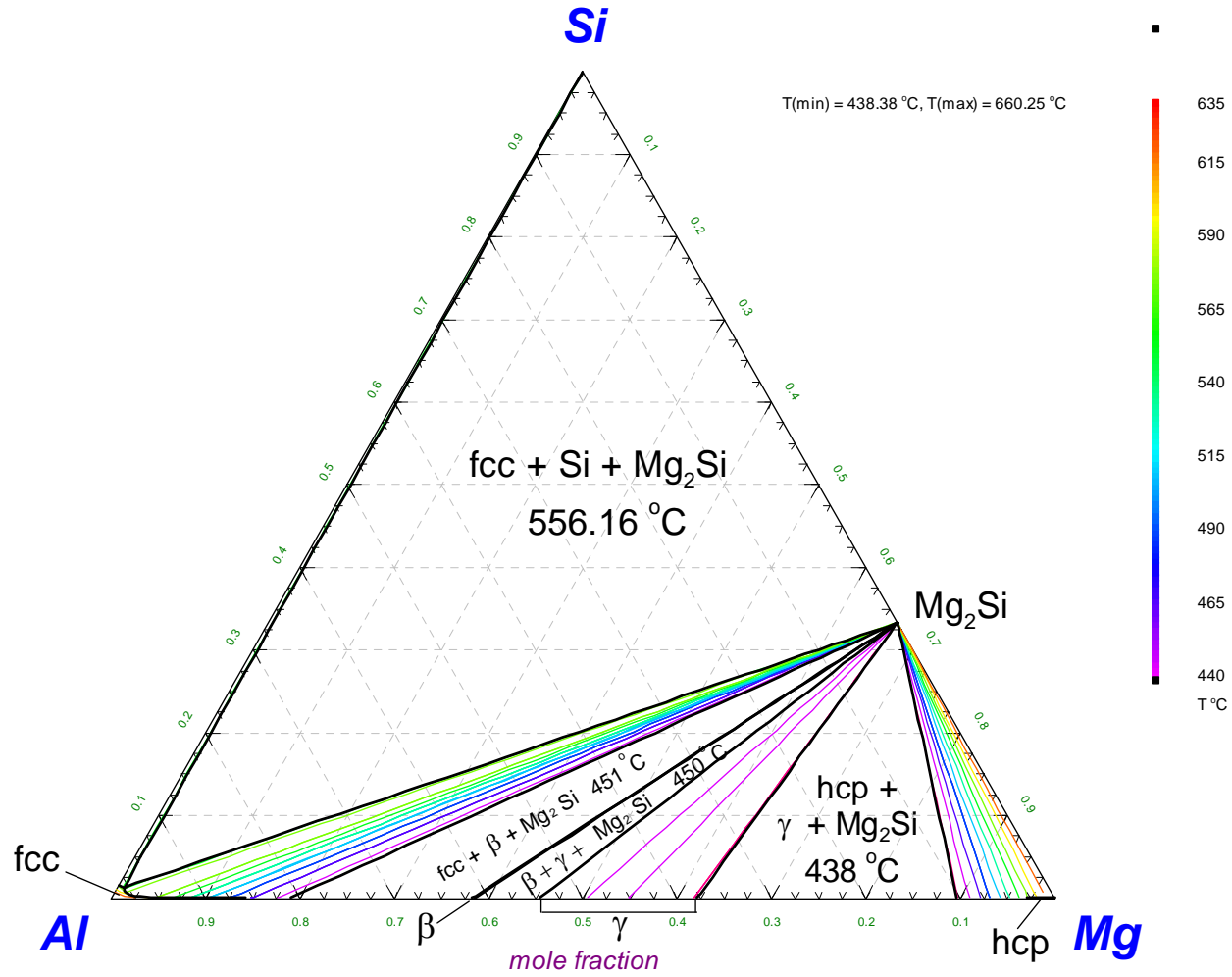


FactSage will calculate phase diagrams with solid projection by the same strategy of following Zero-Phase-Fraction (ZPF) lines as for all other types of phase diagrams.

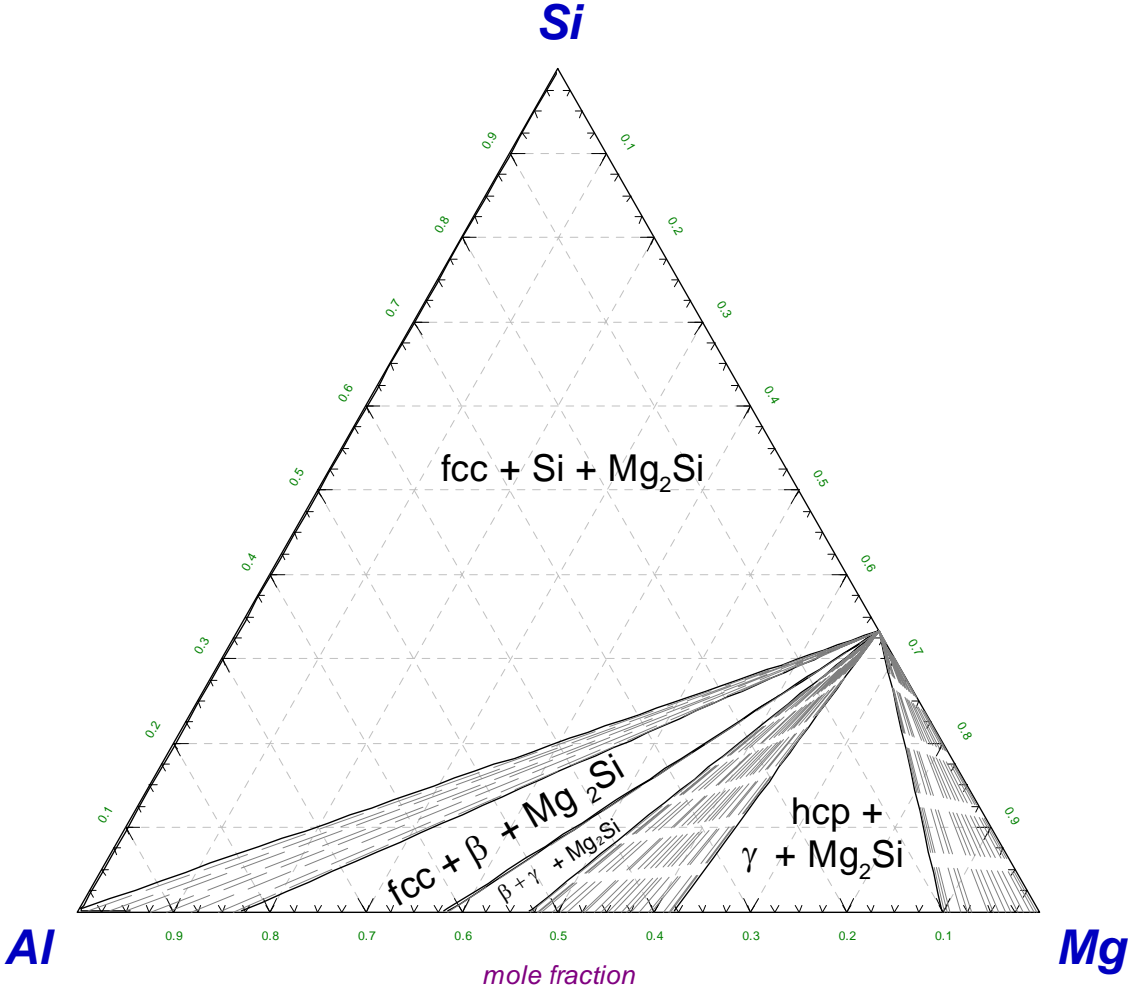
Input parameters for a phase diagram calculation can be chosen such that phase fields might be overlapping. Such phase diagram types are not permitted in FactSage, the reason being that the ZPF-line method will then not work properly.



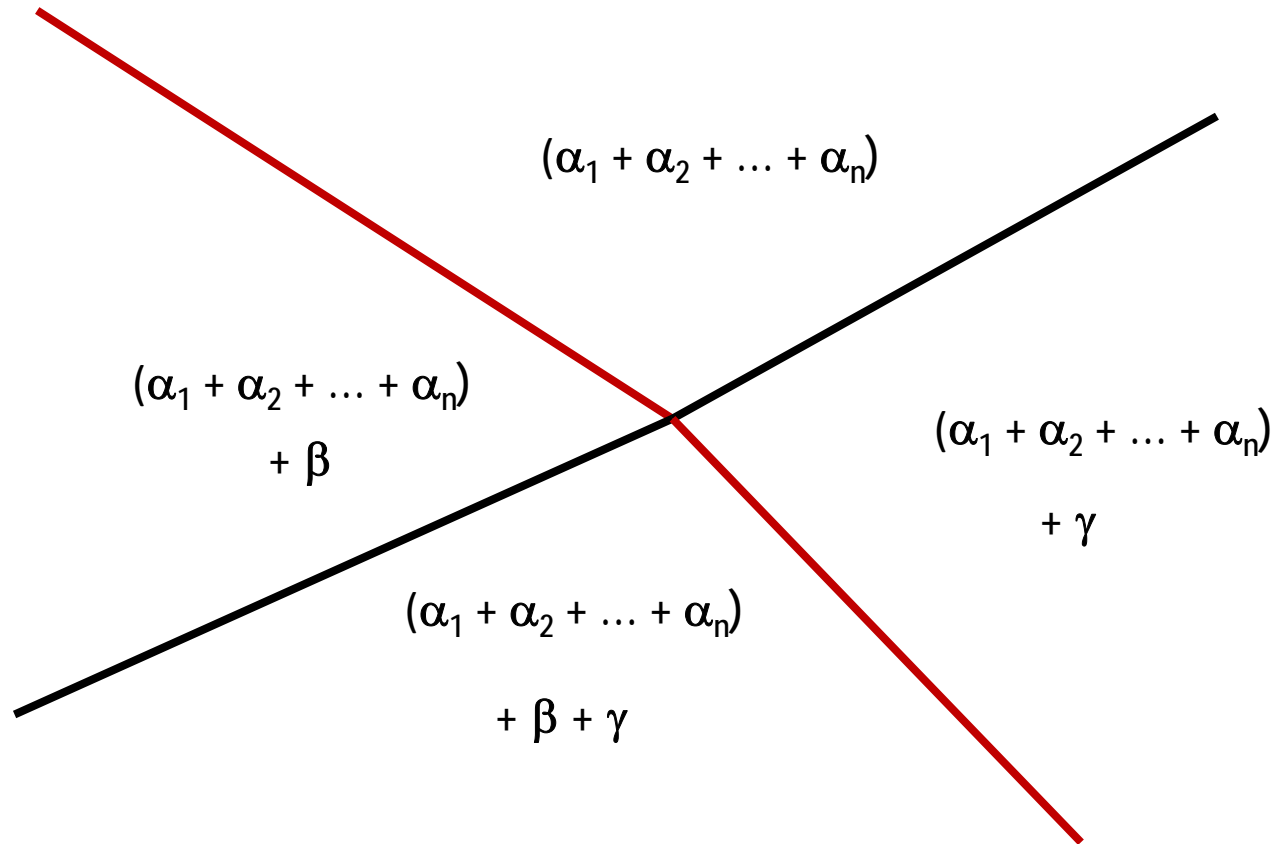
SOLIDUS PROJECTION



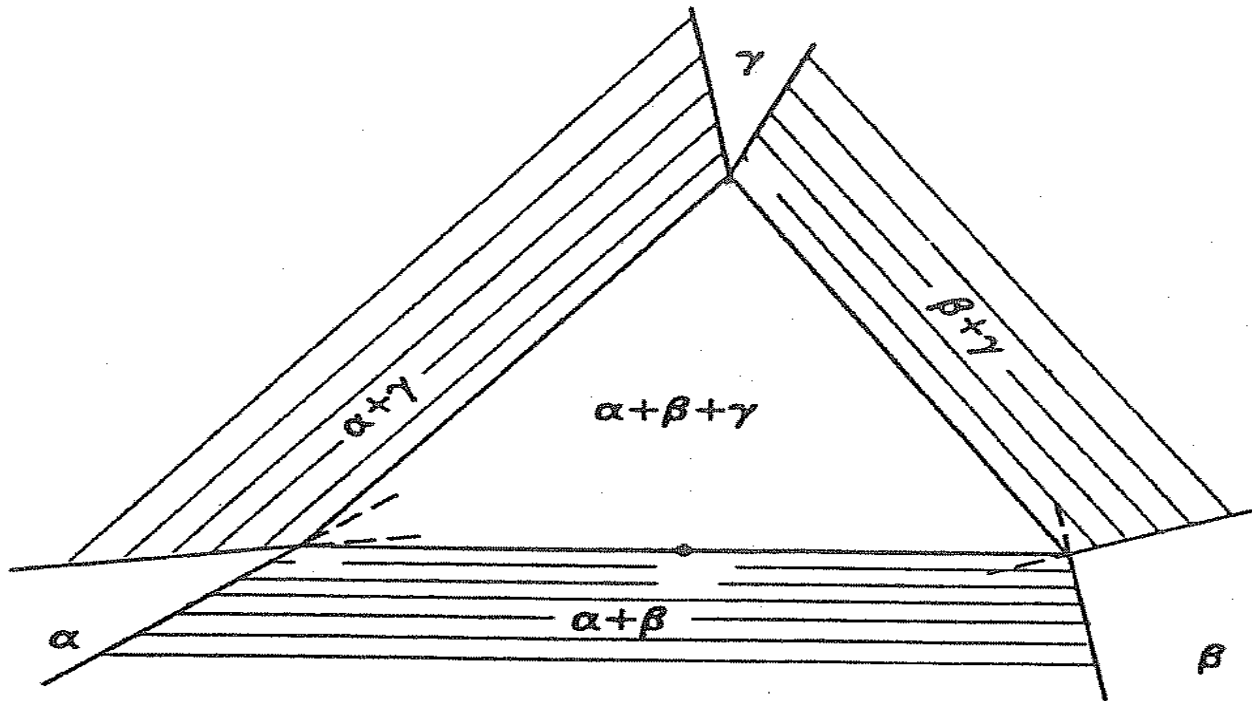
ISOTHERMAL SECTION 430°C



A node in a general phase diagram section or a solidus projection



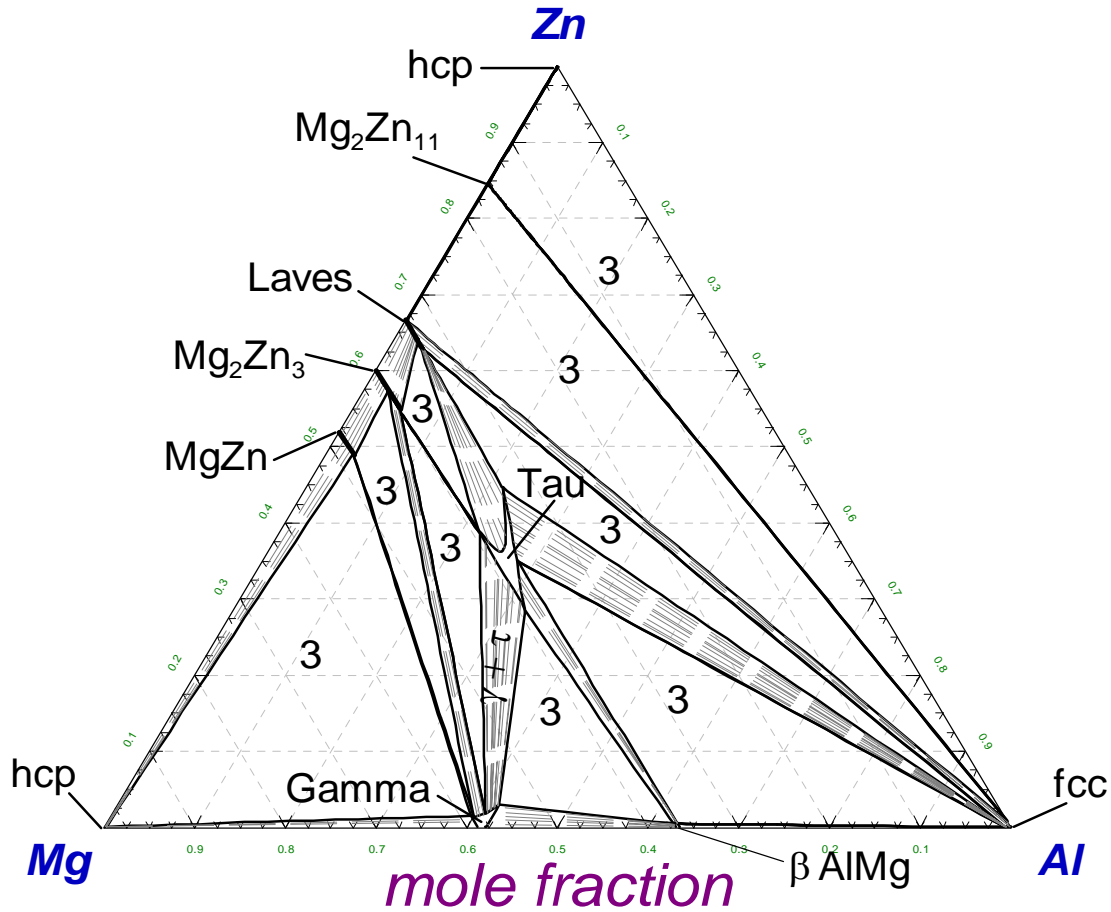
Tie triangle in a ternary isothermal section or in a ternary solidus projection



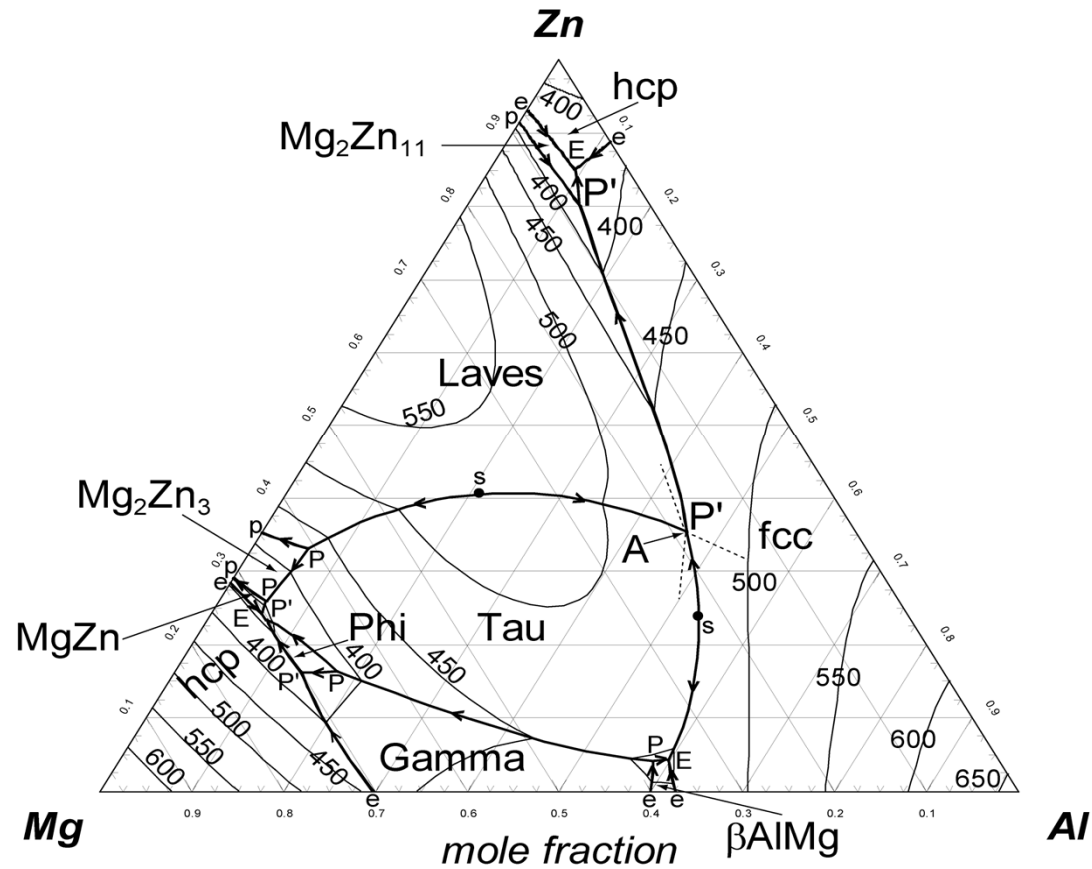
Schreinemakers' Rule



Zn - Mg - Al Isothermal section at 25 °C

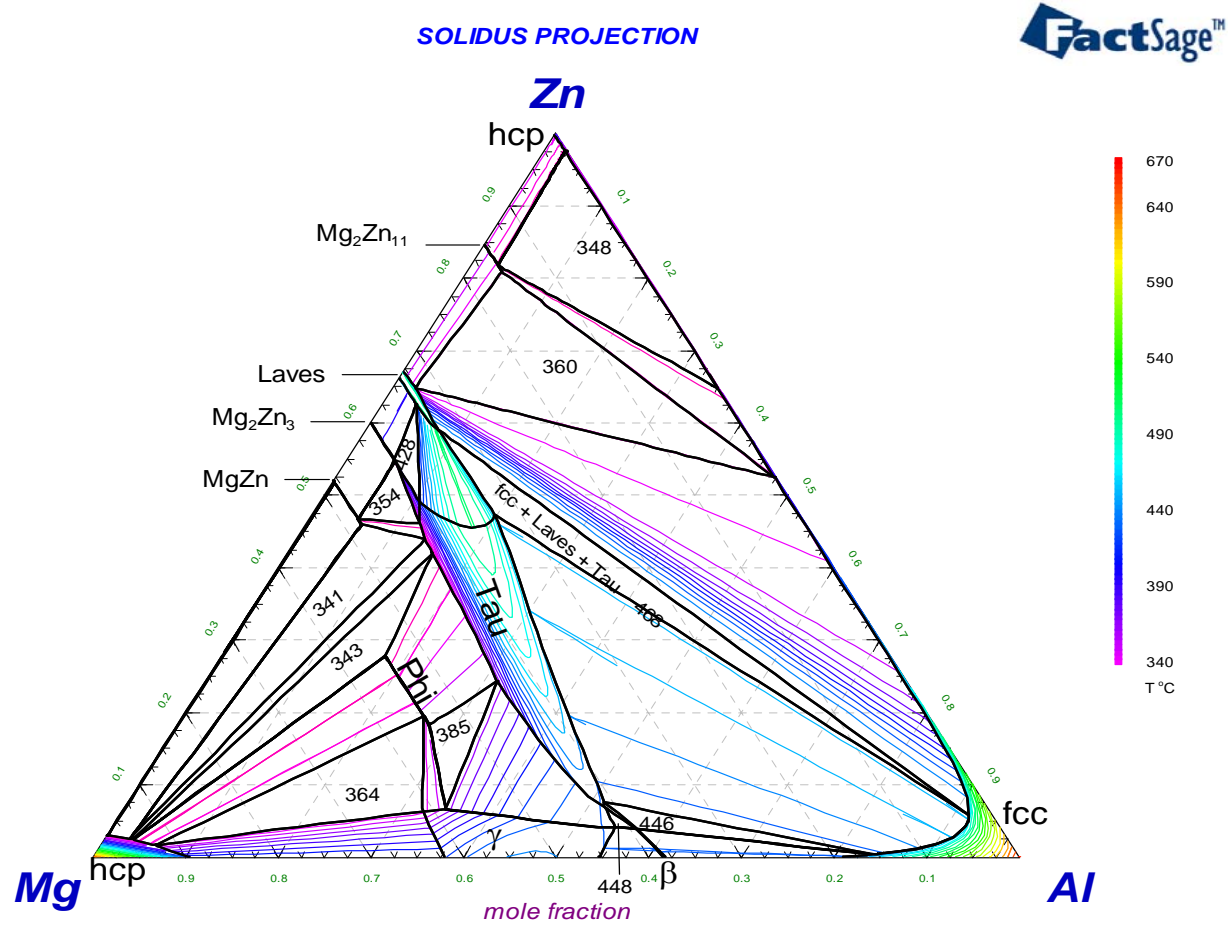


Zn-Mg-Al liquidus projection



Each ternary invariant (P=peritectic, E=eutectic) point on the liquidus projection corresponds to a tie-triangle on the solidus projection

Zn-Mg-Al solidus projection

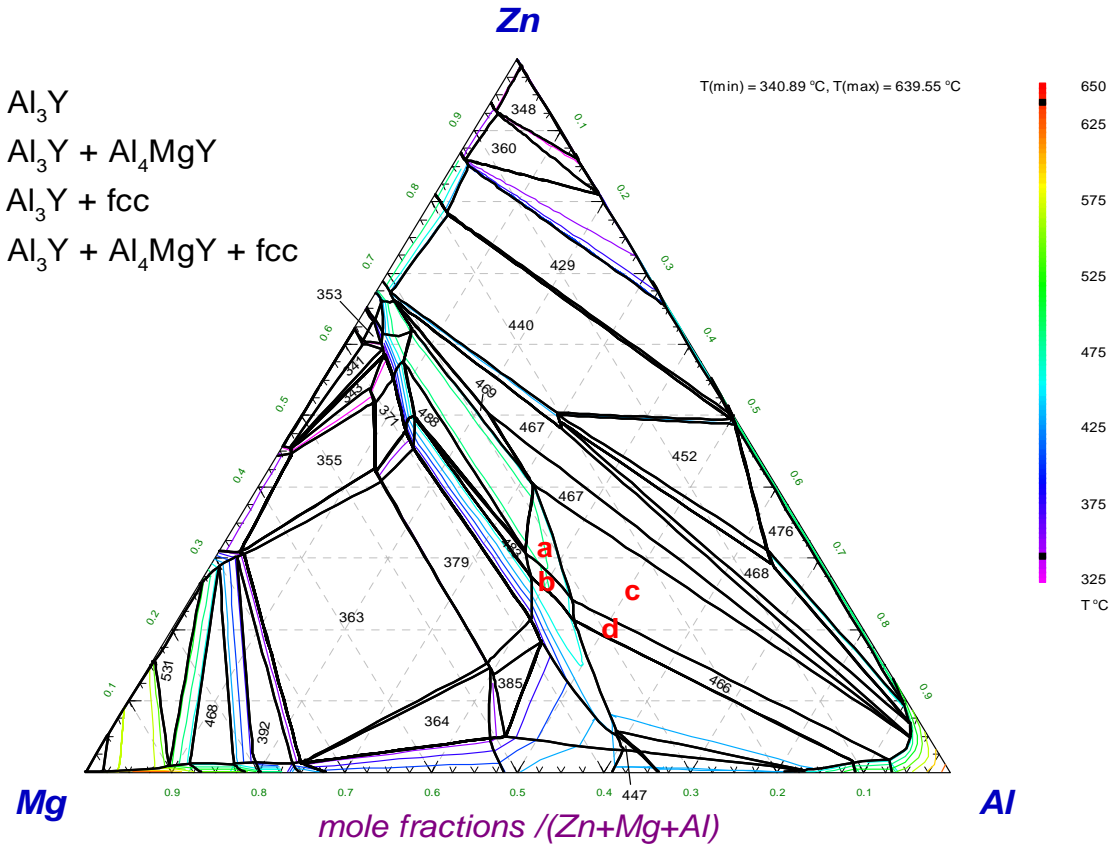


Zn-Mg-Al-Y

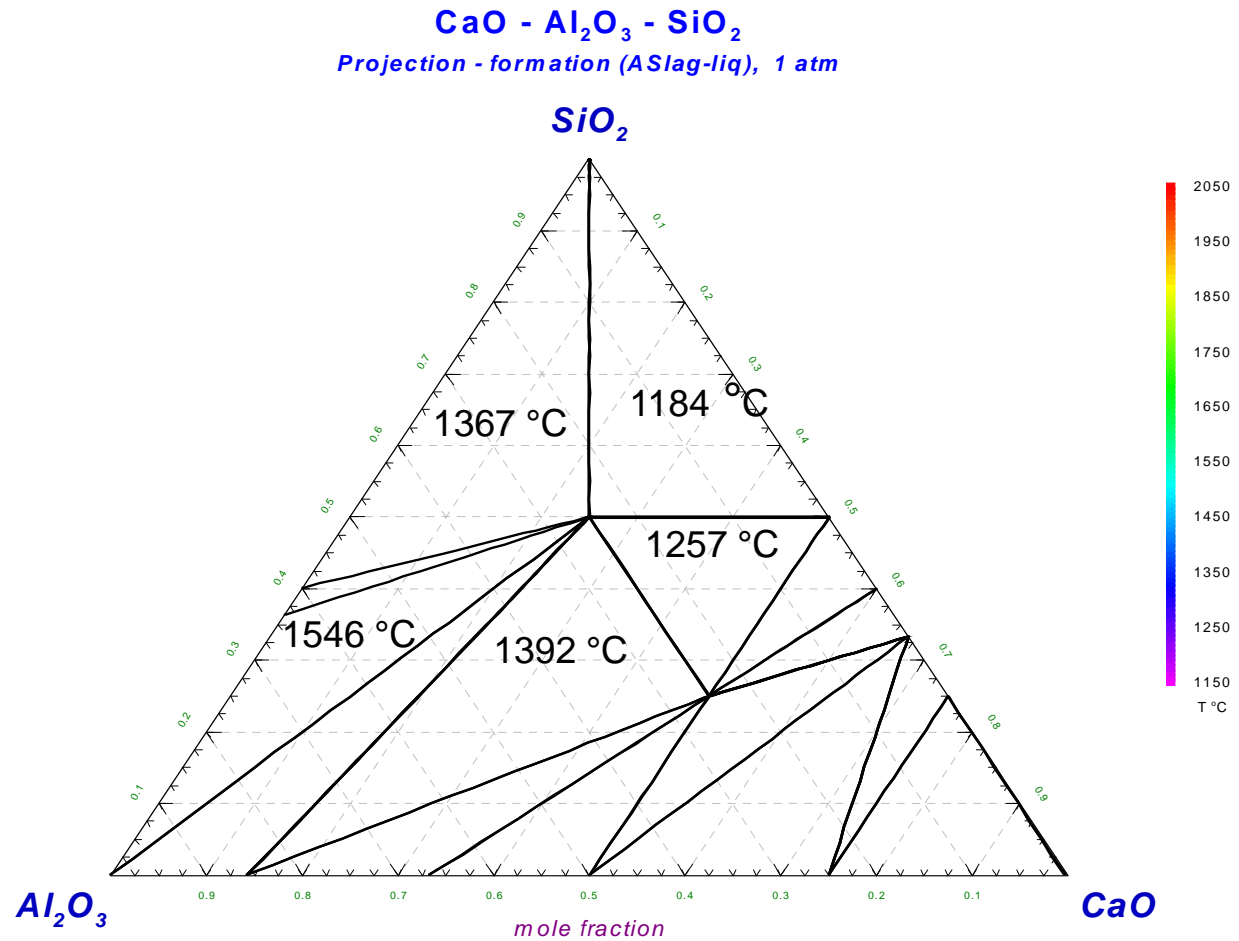
SOLIDUS PROJECTION, mole fraction Yttrium = 0.05



- a = Tau + Al₃Y
- b = Tau + Al₃Y + Al₄MgY
- c = Tau + Al₃Y + fcc
- d = Tau + Al₃Y + Al₄MgY + fcc



Solidus projection when stable ternary compounds



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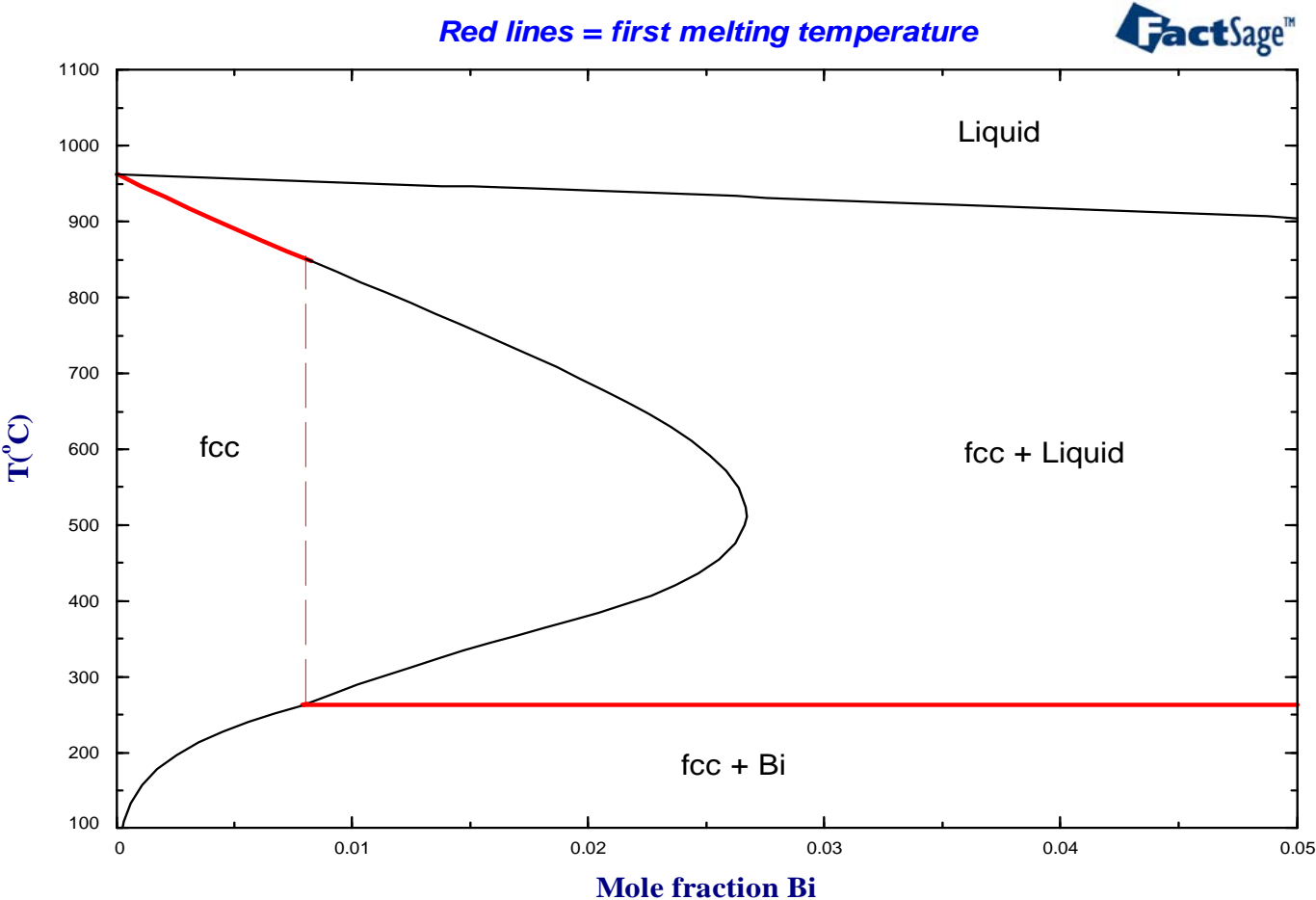


Exceptions and challenges

- In systems with **catatectics** or **retrograde solubilities**, a liquid phase can **resolidify** upon heating.
- In such systems, phase fields on a **solidus projection** can overlap.
- However, phase fields on a **'first-melting-temperature' projection** never overlap. These are calculated automatically by the ZPF strategy.
- If a system contains no catatectics or retrograde solubilities (as is the case in the great majority of systems), the first-melting-temperature and solidus projections are identical.
- **Note:** For systems exhibiting catatectics or retrograde solubility, the usual phase diagram rules are not necessarily obeyed in those composition ranges where liquid resolidifies upon heating.

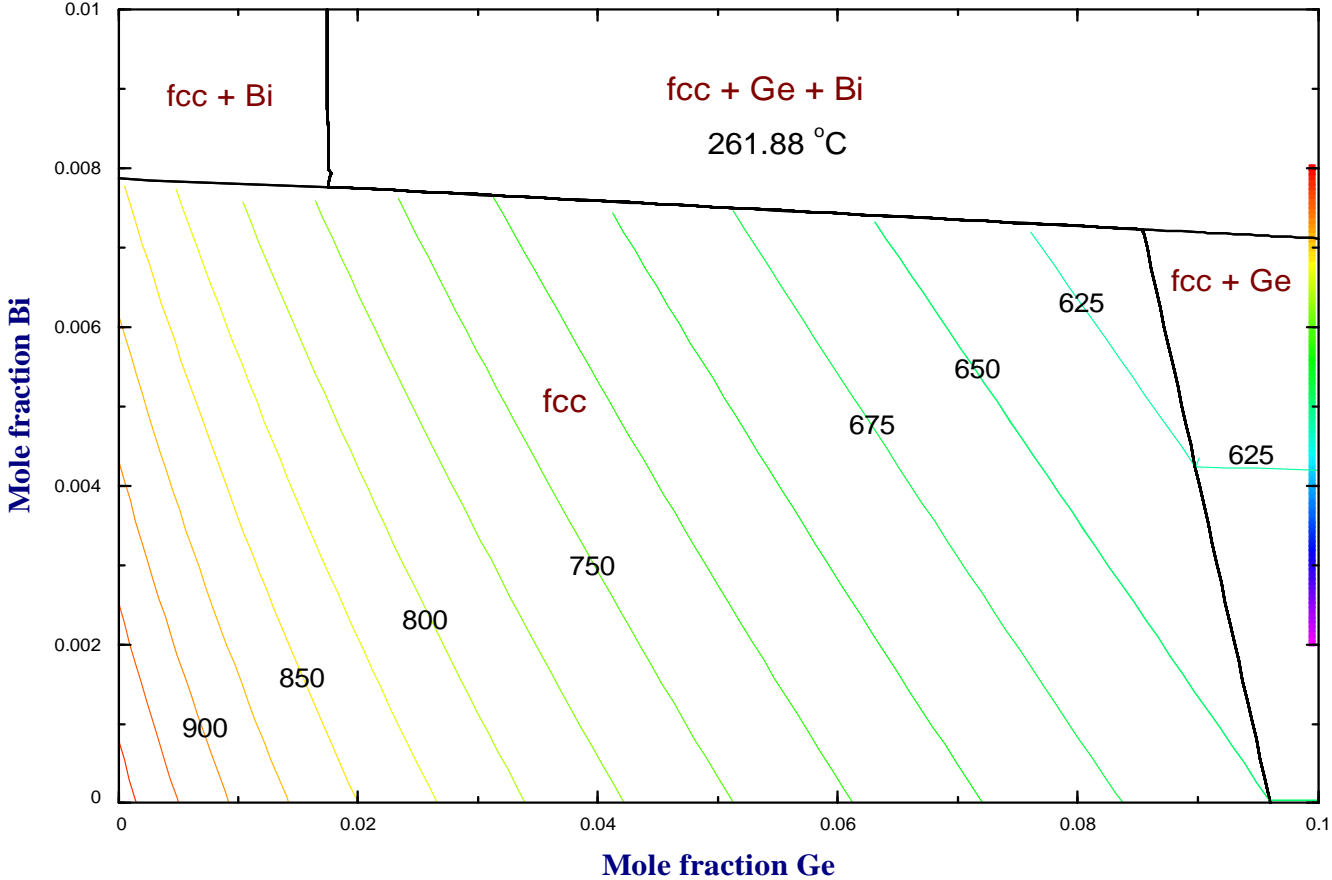


Ag-Bi with a retrograde solubility



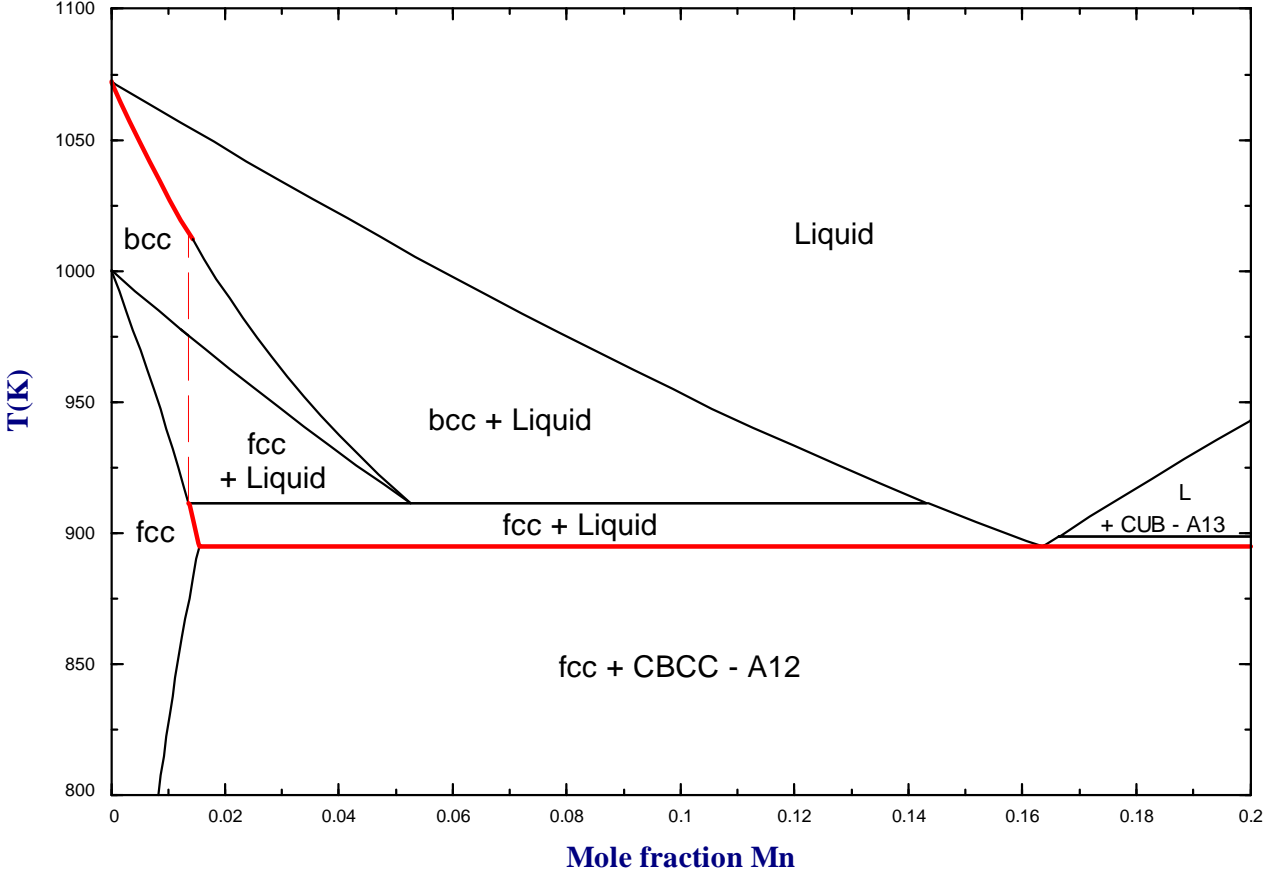
Ag-Bi-Ge

FIRST MELTING PROJECTION

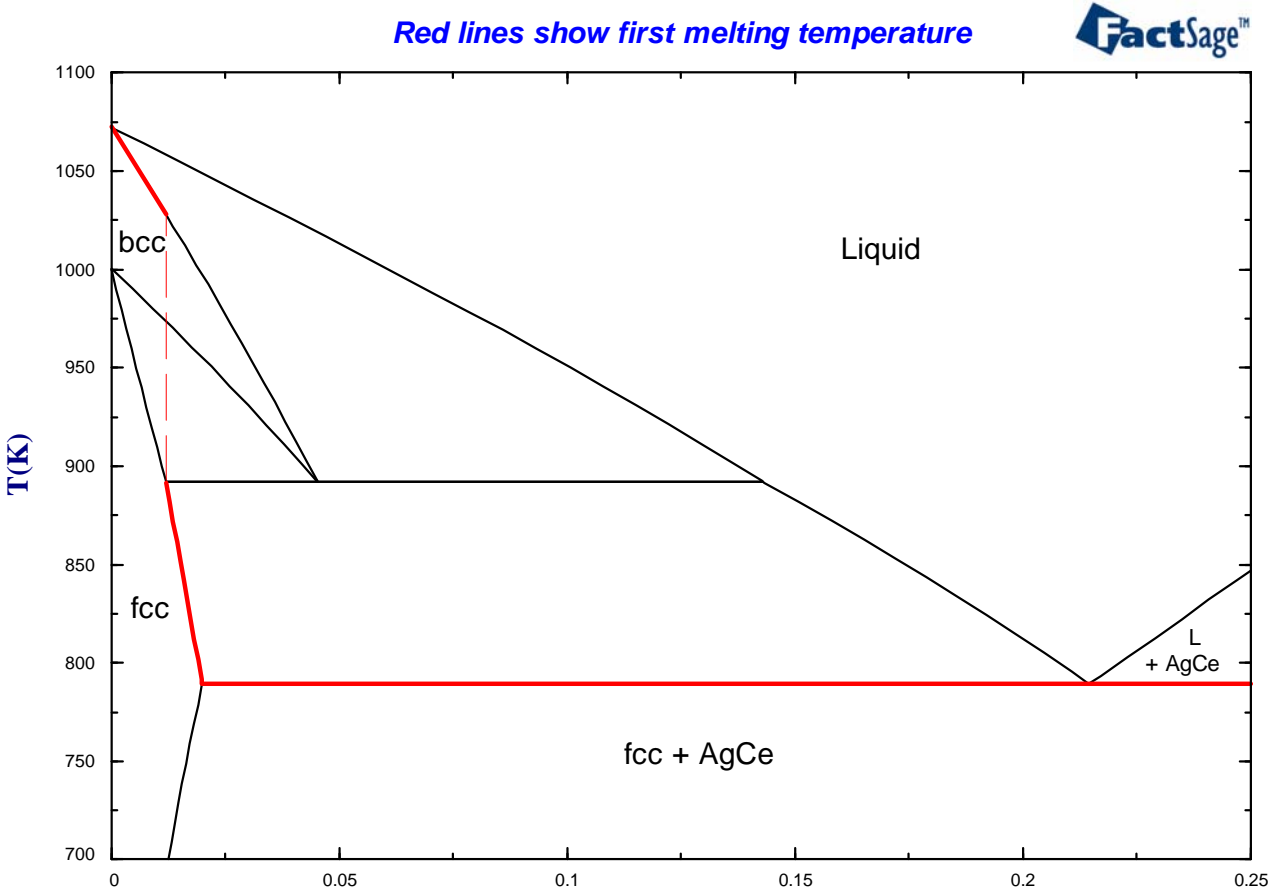


The catatectic system Ce-Mn

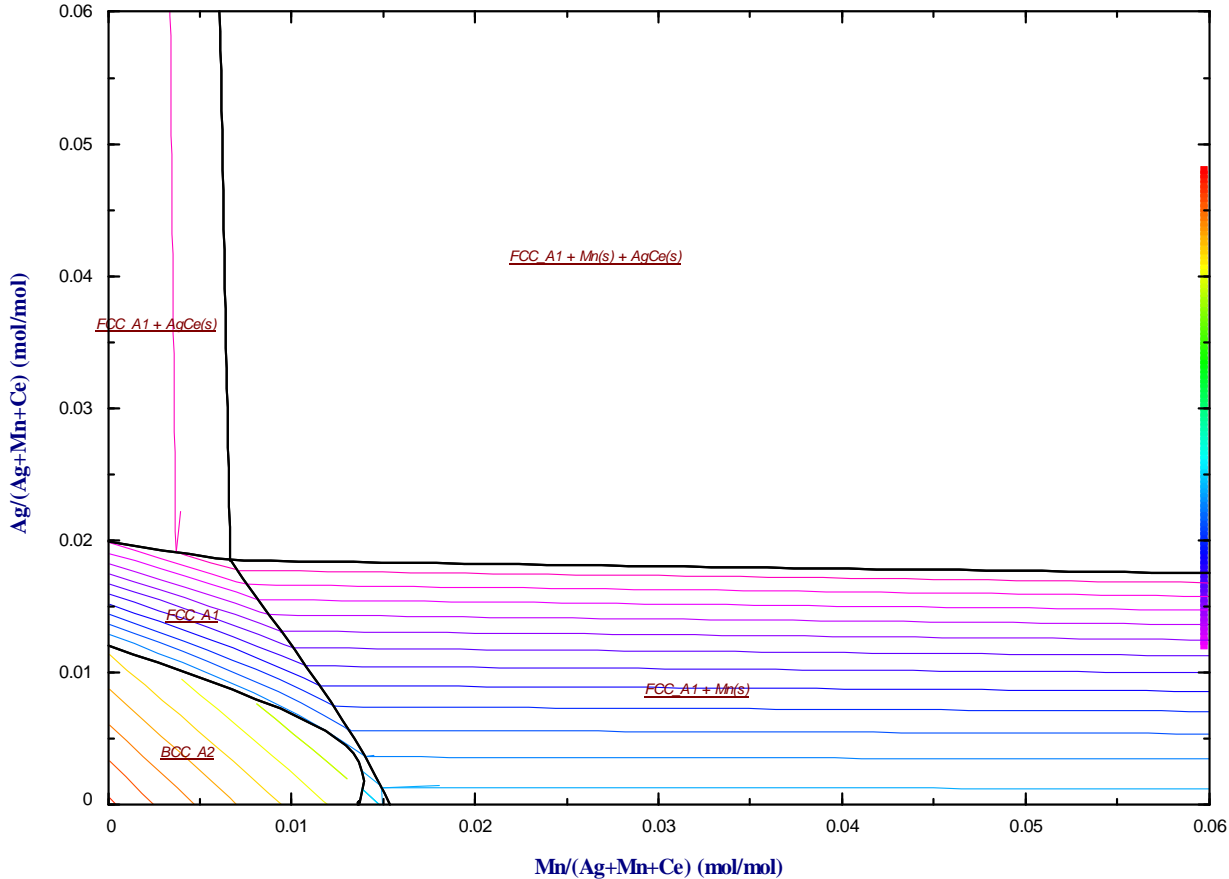
Red lines show first melting temperature



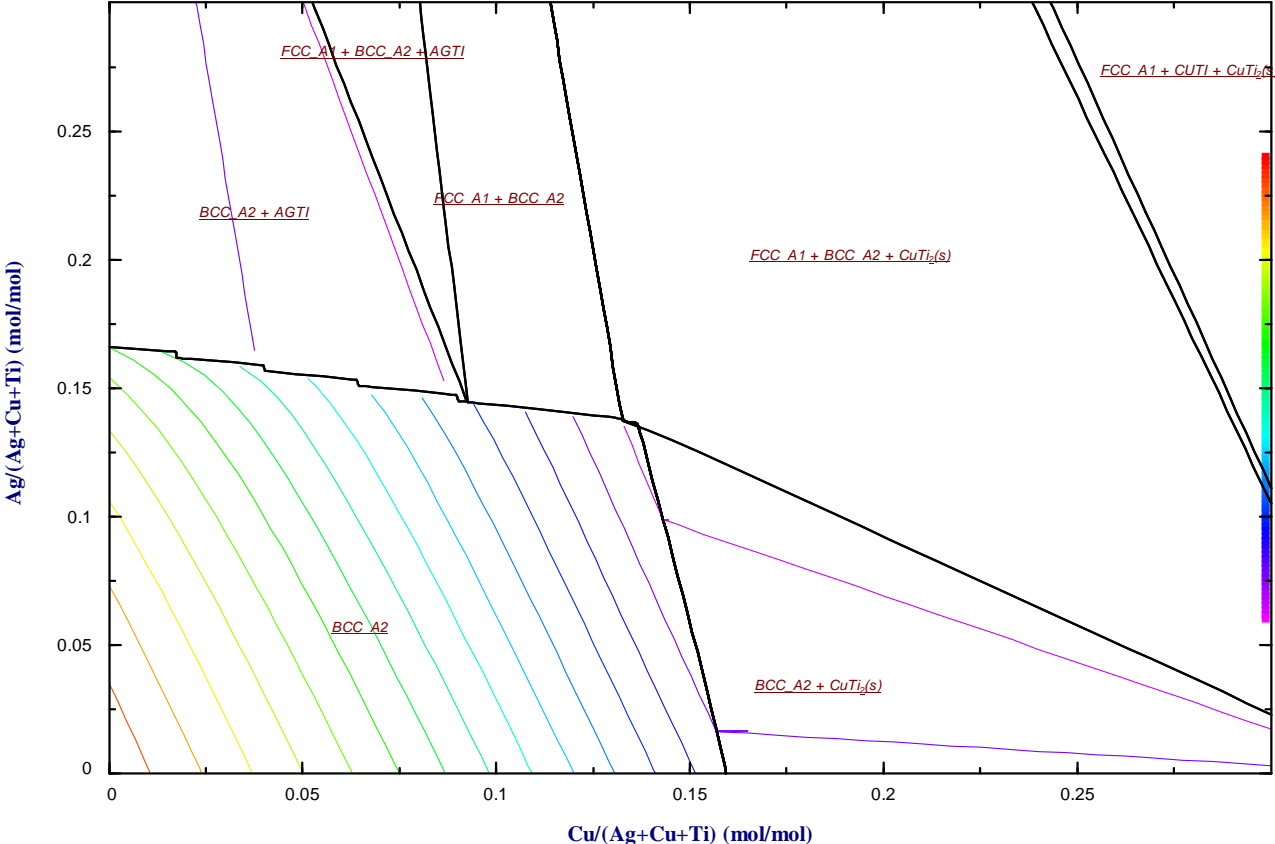
The catatectic system Ce-Ag



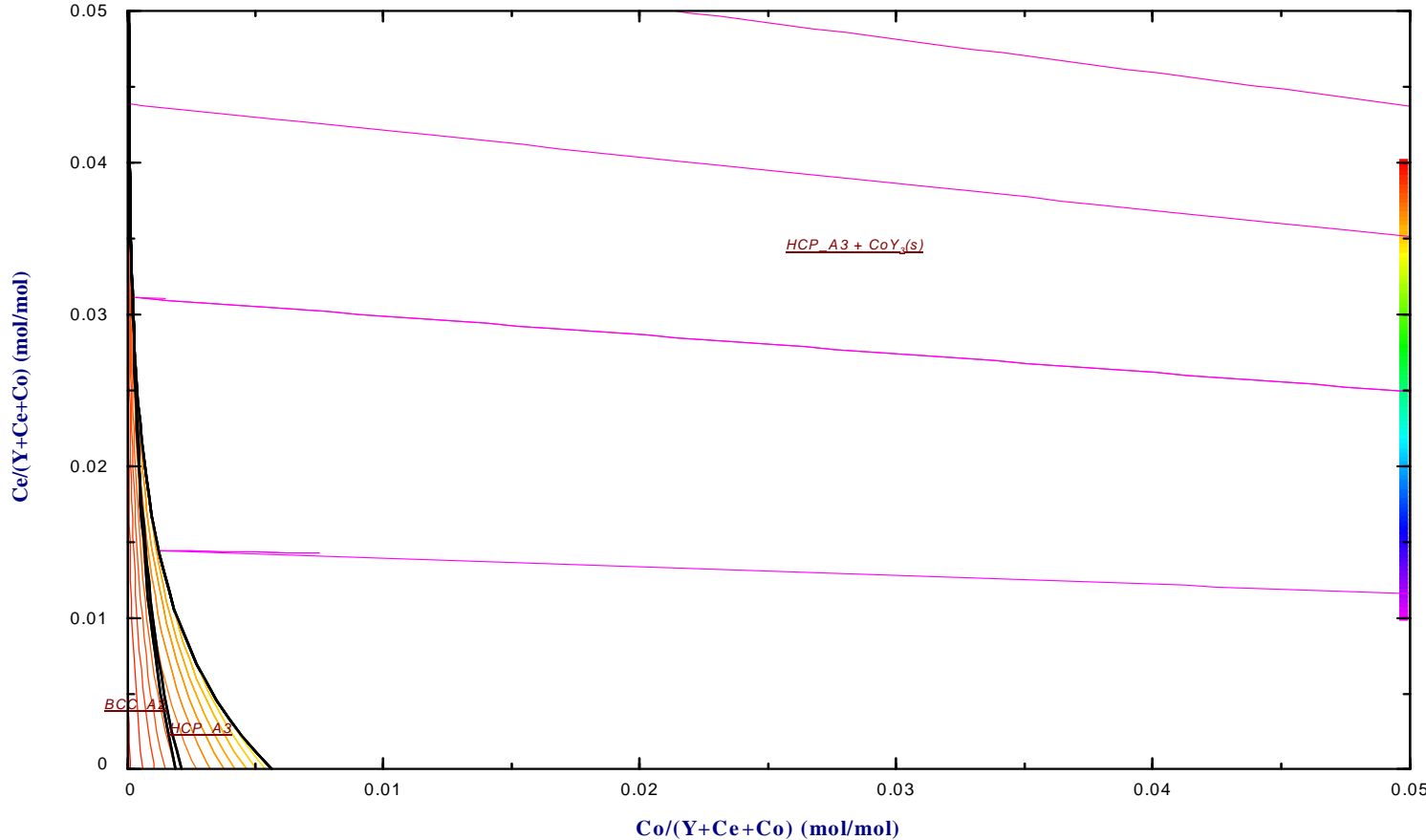
Ag - Mn - Ce
Projection - formation (LIQUID), 1 atm



Ag - Cu - Ti
Projection - formation (LIQUID), 1 atm

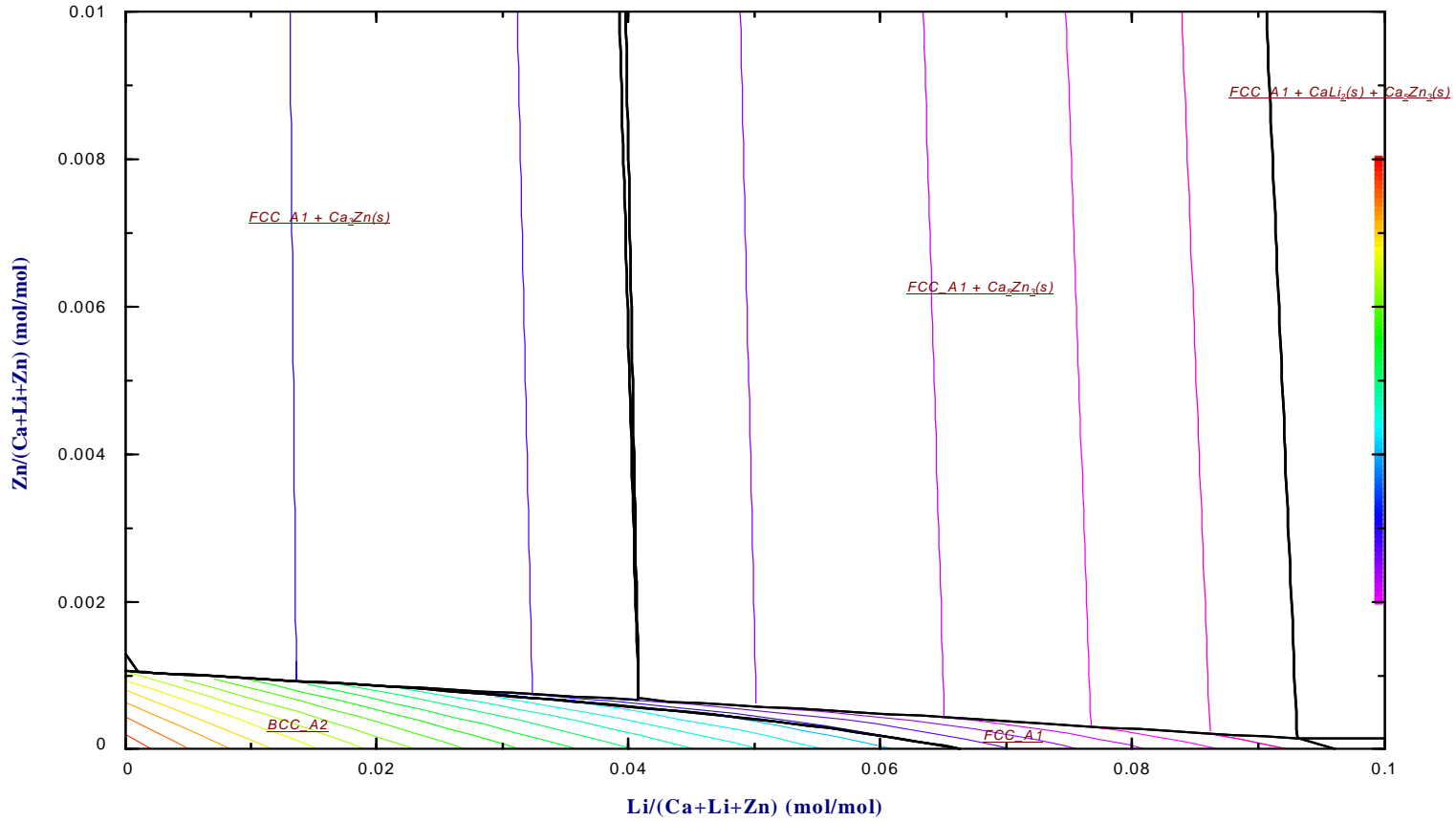


Y - Ce - Co
Projection - formation (LIQUID), 1 atm

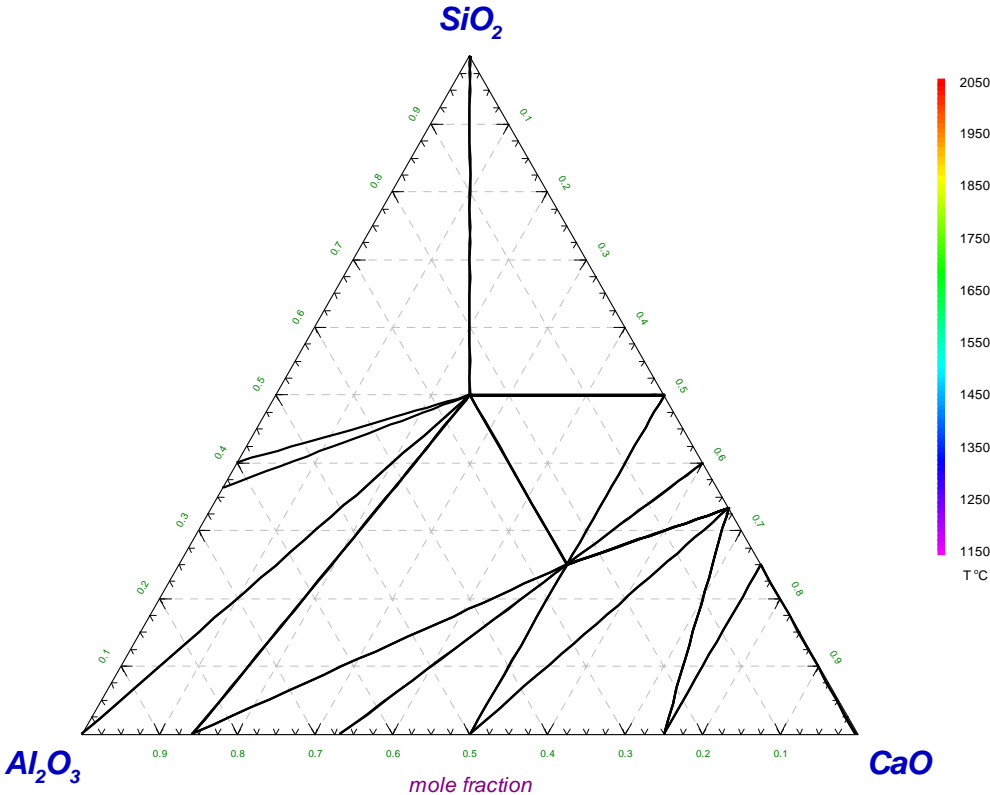


Ca-Zn: retrograde solubility
Ca-Li: catatectics
Li-Zn: not evaluated

Ca - Li - Zn
Projection - formation (LIQUID), 1 atm

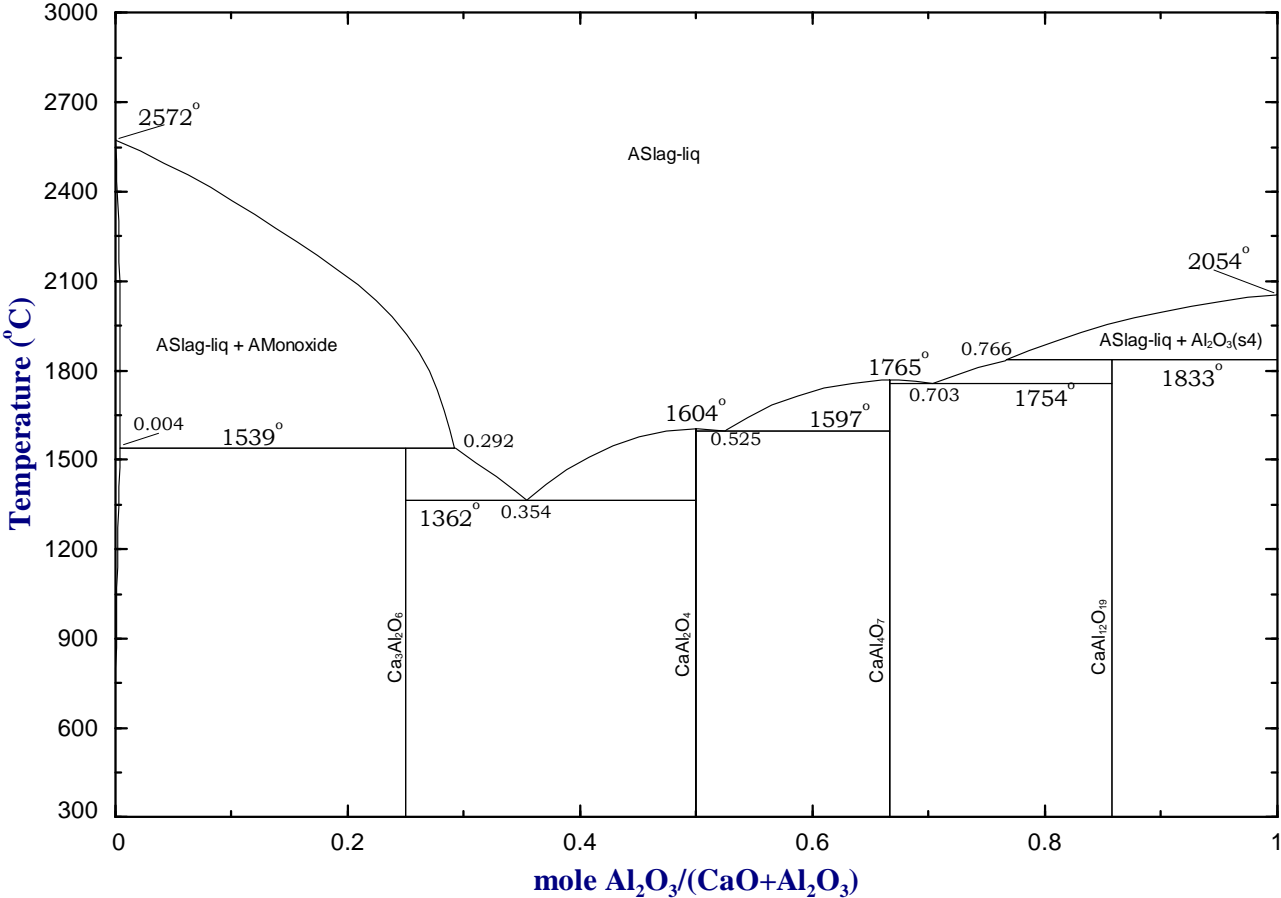


CaO - Al₂O₃ - SiO₂
Projection - formation (ASlag-liq), 1 atm

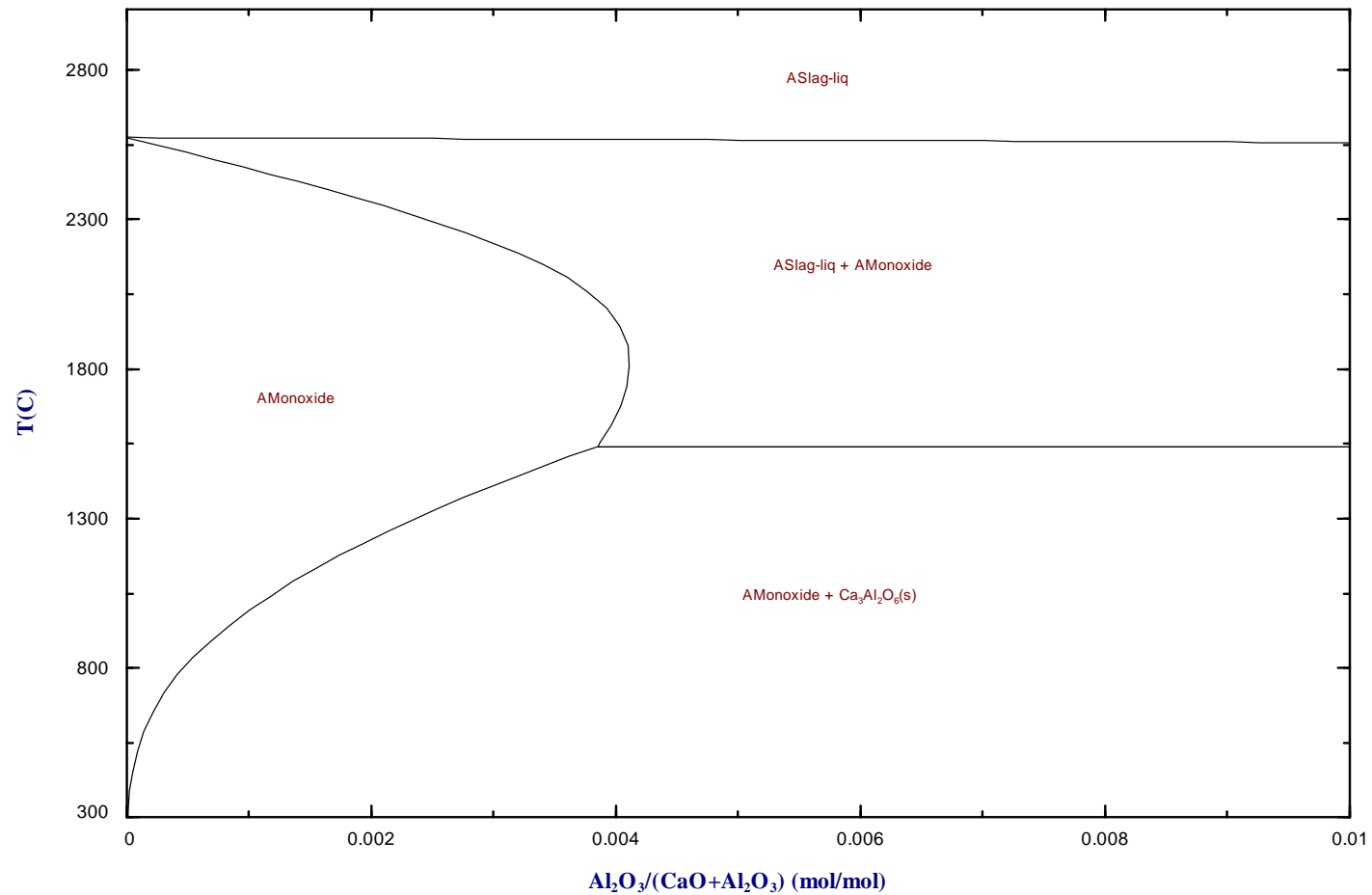


CaO - Al₂O₃

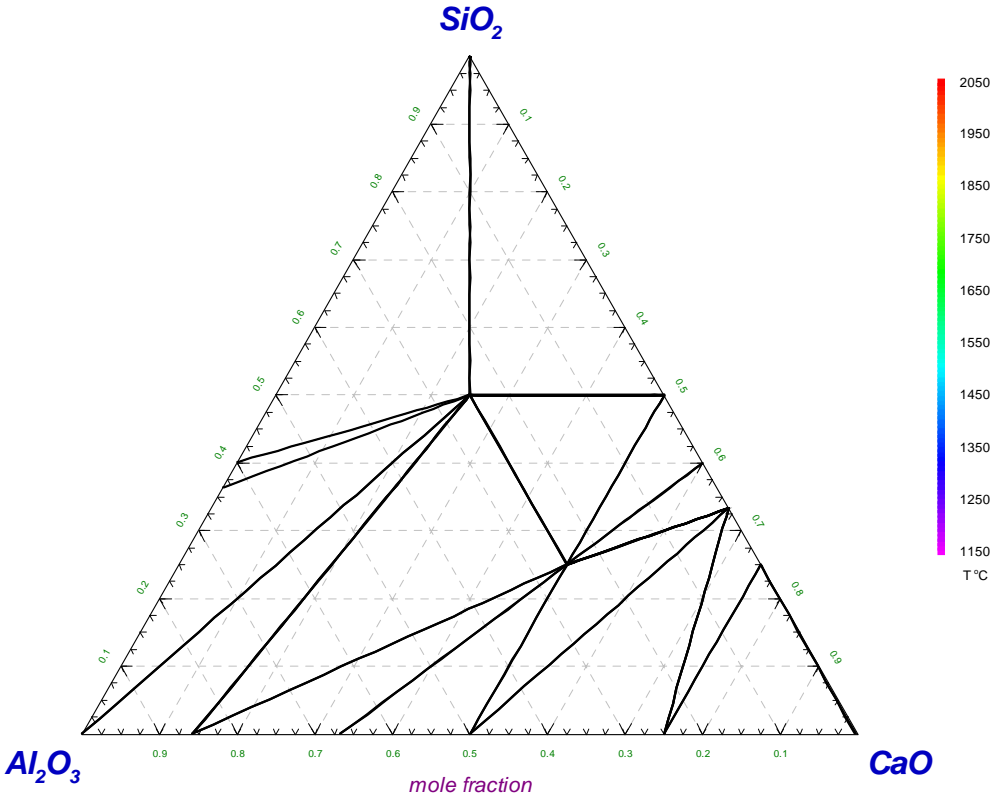
Data from FToxid - FACT oxide database 2010



CaO - Al₂O₃
1 atm



CaO - Al₂O₃ - SiO₂
Projection - formation (ASlag-liq), 1 atm



State-of-the-art solidus vs. first-melting-temperature projection

- Solidus projection works acceptably with FactSage 6.3.
- For solidus projection in FactSage 6.4, temperature would have to be scanned along the phase diagram axes to explore whether or not the system to be calculated exhibits retrograde solubilities or catatectics. When this is found to be the case, a first-melting-temperature projection would have to be executed.



Thank you for your kind attention!

